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ORIGINAL RESEARCH

Formulation and evaluation of a topical niosomal gel containing a combination of benzoyl peroxide and tretinoin for antiacne activity

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d normally Abstract: A skin disease, like acne, is very common everyone at least pens once in their lifetime. The structure of the stratum ameum justen constructed with a brick wall, with corneocytes surrounded by the mortar of alar lipichamellae. One of the best the options for successful drug delivery to the skin is e use of elastic vesicles (niolected area el-like structures. In this study, a the sin through c. somes) which can be transported through combination of tretinoin (keratolytic agent) and nzoyl peroxide (BPO) (a potent antibacterial) was given by using niosomes the effective treatment of acne by acting Ising carriers on a pathogenic site. In this section, niosomal gel formulation encapsulated drugs have been evaluated for in vitro, ex vi , and in vivo or their predetermined characteristics; and finally the stability of the niosome was tested t different temperature conditions for understandditions requ ing of the storag maintaining the quality of formulation attributes. The prepared niosome be in the range of 531 nm with a zeta potential of -43 mV; the as fou. entrapment efficientes retinom (TRA) and BPO niosomes were found to be 96.25%±0.56% .25%, pectively. The permeated amount of TRA and BPO from the niosomal and 9 0% urs was after 24 culated as $6.25\pm0.14 \,\mu\text{g/cm}^2$ and $5.04\pm0.014 \,\mu\text{g/cm}^2$, respectively. A tion study in Wistar rat skin using cream, an alcoholic solution, and a arat gel showed 11.54 µg, 2.68 µg, and 15.54 µg amounts of TRA and 68.85 µg, 59.98 µg, niosol g amounts of BPO were retained in the layers of skin, respectively. In vivo studies and 143. the niosomal gel and antiacne cream of TRA and BPO showed that the niosomal gel was more acious than the antiacne cream because niosomal gels with a 4.16-fold lower dose of BPO provided the same therapeutic index at targeted sites in comparison to the antiacne cream. Keywords: antiacne combination therapy, rabbit ear pinna model, retention efficiency, therapeutic index

Introduction

The optimization of drug delivery through human skin is important in modern therapy. Clearly, the topical route of drug delivery for treating skin diseases offers an attractive alternative to the conventional drug-delivery methods of oral administration/injection, and it is becoming a most innovative research area in drug delivery. A skin disease like acne, is very common and normally happens to everyone once in their lifetime.

Acne vulgaris is a chronic inflammatory dermatosis which is notable for open and/or closed comedones (blackheads and whiteheads), and inflammatory lesions including papules, pustules, or nodules. It is a disorder of sebaceous follicles which are special pilosebaceous units located on the face, chest, and back.¹ Propionibacterium acnes and Staphylococcus epidermidis have been recognized as pus-forming bacteria triggering inflammation in acne.²

The organism produces extracellular lipases that hydrolyze sebum triglycerides to glycerol and free fatty acids that have proinflammatory properties.³ The topical treatment of acne

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includes topical retinoids,^{4,5} benzoyl peroxide (BPO),^{4,6} azelaic acid,⁷ erythromycin,⁸ clindamycin,⁸ and combination therapies.^{9,10} The adverse effects of topical antiacne agents include burning, erythema, scaling, flare-up, photosensitivity, and bacterial resistance.4 Tretinoin (TRA) and BPO are used individually and in a cyclic manner for acne treatment. Various conventional topical medicines are available in the market for treatment but have a less-therapeutic effect due to the efficient barrier properties of skin membranes. The structure of the stratum corneum is often compared with a brick wall made of corneocytes and surrounded by the mortar of the intercellular lipid lamellae.¹¹ The best alternative for successful drug delivery to an affected area of skin is elastic vesicles (niosomes) which can be transported through the skin via channel-like structures. Moreover, they are too small-in the nanometer size range - to be detected by the immune system; furthermore, they can deliver the drug to the target site using lower drug doses in order to reduce side effects often experienced by topical routes by passing the complexity of the skin structure.¹² The main advantages of using nanocarriers arise from their peculiar features, such as their tiny size, high surface energy, high surface area, composition, and architecture.¹²

Colloidal particulate carriers (including niosomes and liposomes) can act as drug reservoirs.¹³ Niosomes are unilamellar or multilamellar nonionic surfactant vesicles formed fro synthetic nonionic surfactants by hydration, offering an alterna tive to liposomes. Niosomes are advantageous from hnical point of view as they possess greater stability a . avoic ome disadvantages associated with liposomes such variab of phospholipids and high cost.¹⁴ The partic c carriers have been extensively studied as dry parriers in to, al drug delivery. These carriers are advant geous ause they increase drug stability, enhance therapeatic effects, prong circulation time in a biological envir ament, and promote the uptake of the entrapped drugs into be tar of site while drug toxicity is educt, in nons cific tissue uptake.¹⁵ diminished due to Niosomes are apable Is ng both hydrophilic and f encap lipophilic days and rve as effective drug carriers.¹⁶

The vesicle verse as a soluble matrix and also serve as a local depot for sust ined drug release; permeation enhancers of dermally active compounds; or a rate-limiting membrane barrier for the modulation of systemic absorption of drugs via dermal drug delivery. Here, Span 60 and cholesterol were selected as components of niosomes with BPO and tretinoin as model drugs for niosomal formulation. An in vitro permeation and retention study of BPO and tretinoin from niosomal gels were performed. Comparative antiacne activity of cream and niosomal gels was further evaluated by using a rabbit ear pinna model and we evaluated the impact of the niosome vesicle in drug delivery at the targeted site.^{17,18}

Tretinoin was procured ex gratis from Shalaks Pharmaceutical, New Delhi, India, while BPO was gifted by the H.K. Group, Mumbai, India. Span 60 and oleic acid were obtained from SD Fine Chemicals Limited, Mumbai, India, and cholesterol was obtained from Central Drug House (P) Ltd, New Delhi, India. Isopropanol was obtained from the Central Drug House (P) Ltd, carbopol 934 from Hi-media Laboratories PVT Ltd, Mumbai, India, and phosphotungstic acid from the Central Drug House (P) Ltd, reacting there materials and chemicals were of analytical grade.

Incompatibility studies between dugs

Incompatibility studies tween dags were afformed using , equal counts of each drug a stability chamber this trans fred to light resistant were taken, mix q uniform d in a stable y chamber. Individually, glass vials, z a pla each drug was also placed at 65% relative humidity and 45°C ature for 1 month. Frared and ultraviolet (UV) spectemp bpy were us to investigate any interaction between the tros (Shimadzu orporation, Kyoto, Japan, and Systronic, drug dia: M del 2201, respectively). Gujrat,

Susceptibility testing of BPO against S. epidermidis

asceptibility testing of *S. epidermidis* against BPO was checked by the disc diffusion method. *S. epidermidis* was incubated in a nutrient agar medium for 24 hours at 37°C and adjusted to yield approximately 1.0×10^8 colony forming units/mL. The prepared inoculums were added to molten agar, mixed, poured over the surface of the agar base, and left to solidify. A sterile paper disc was impregnated with test material and the disc was placed on the agar plates. The plates were incubated at 37°C for 24 hours under aerobic conditions. All disc diffusion tests were performed in three separate experiments and the antibacterial activity was expressed as the mean of the inhibition diameters (mm).

Minimum inhibitory concentration of BPO for S. epidermidis

The minimum inhibitory concentration (MIC) value was determined by the microdilution or broth dilution method. The calculated amount of broth was measured in a 10-mL test tube and the glass test tube was sterilized by autoclaving at 121°C for 15 minutes. The medium was cooled and inoculated with 100 μ L of a bacterial suspension containing 10⁸ cells/mL. Then, various concentrations of BPO were added to respective test tubes which were incubated at 37°C±1°C for 24 hours aerobically, and the

growth of *S. epidermidis* was measured as function of turbidity at 660 nm using a UV spectrophotometer (Systronic 2201).¹⁹

Preparation of niosomes

An accurately weighed nonionic surfactant (Span 60) and cholesterol were dissolved in a chloroform:methanol (2:1) mixture and placed into a round bottom flask. The required quantity of stock solution of tretinoin (4 mg/mL) and BPO (15 mg/mL) were added in an optimized surfactant:cholesterol ratio as per batch size, then the organic solvent was removed by applying a vacuum. The temperature of the bath was set at 60°C and the flask was rotated at 160 rpm until a smooth film was formed. Film was removed from the round bottom flask using a rotary evaporator equipment and put aside for 12 hours to remove traces of an organic solvent. Then, hydration of the film was performed with an optimized volume of water and saline, (in present study, water is used as hydration media for preparation of BPO niosomes, and saline is used for preparation of tretinoin niosomes), at above the lipidtransition temperature of the surfactant. Niosomes were formed and observed under a microscope (Tables 1 and 2).20

Characterization of niosomes Morphological analysis by transmission electron microscopy

bon A drop of diluted niosome dispersion was applied to a ca coated 300-mesh copper grid and was left for 1 minut to allow for some of the niosomes to adhe 10 e carb substrate and be stained with 1% phosp¹ stungsti acid. Th remaining dispersion was removed by Sor ng the with the corner of a piece of filter aper. The samples were examined and photographed with Hitachi L (Tokyo, e at 100 KV.21 Japan) transmission electron microse

Particle size analysis by photon correlation spectroscopy (dynamic aser light scattering [DLLS]) The vesicle sizes emission as were elermined by light scattering based or aser diffection using a Malvern Mastersizer (Model S, Ver. 205; Marven, C, aments, Malvern, UK). The apparatus consists of a HeNe laser (5 mW) and a small-volume sample-holding cell. The sample was stirred using a magnetic stirrer bead to keep and maintain the sample in suspension.

Zeta potential

The significance of zeta potential is that its value can be related to the stability of colloidal dispersions. The zeta potential indicates the degree of repulsion between adjacent, similarly charged particles in dispersion. The zeta potential for the niosomal dispersion was determined using Malvern instruments.²²

Encapsulation efficiency

Niosome-entrapped BPO and tretigent courses be separated from the free drug by the dialysis method. After hydra n step, suspension of niosomes will form which as filled in alysis bags for removal of free drug present in suspenting (M CO-14000) and the free drugs were dyzed fr 24 hours to 100 mL of a phospH 7.4), ter 24 hours, the dialysis phate buffer salie solu ed *y* the entrapped drug. From niosomal sy ension cont. pension, 0.5 nL was taken and isopropanol this nior nal s. was added up to 5 n, then the volume was increased to 10 mL the respective solvent (tretinoin niosomal suspension with nethanol and PO suspension with ethanol); then, the absornce of the resulting solution was measured at 234.8 nm and BPO and tretinoin, respectively.²³

Dn. ential scanning calorimetry analysis for determining the phase transition temperature of niosomes (glass transition temperature)

Differential scanning calorimetry (DSC) experiments were performed with a differential scanning calorimeter (Shimadzu Corporation; model TA-50 WSI) calibrated with indium. Samples of multilamellar niosomes composed of Span 60:cholesterol (207:52 or 138:52 mg ratio) were submitted for DSC analysis. The analysis was performed on 40- μ L samples sealed in standard aluminum pans. Thermograms were obtained at a scanning rate of 10°C/minute. Zero point nine percent saline was employed as a reference. Samples were scanned between 30°C and 300°C. The maximal excess heat capacity was defined as the phase transition temperature.^{24,25}

Table I Ratio of surfactant and cholesterol used for niosome preparation, percent encapsulation of benzoyl peroxide

Serial	Niosomal formulation loaded with benzoyl peroxide	Hydration media	% encapsulation
number			efficiency (SD)
I	Span 60:CH (69:35) weight (mg) ratio	Water	52.6%±0.45%
2	Span 60:CH (69:35) weight (mg) ratio	Saline	51.1%±0.34%
3	Span 60:CH (138:52) weight (mg) ratio	Water	98.75%±1.25%
4	Span 60:CH (138:52) weight (mg) ratio	Saline	94.86%±0.56%
5	Span 60:CH (207:52) weight (mg) ratio	Water	92.4%±0.49%
6	Span 60:CH (207:52) weight (mg) ratio	Saline	89.06%±0.76%

Notes: Stock solution of benzoyl peroxide was 15 mg/mL. Percent encapsulation efficiency is the mean from triplicate experiments. Abbreviations: CH, cholesterol; SD, standard deviation.

Serial number	Niosomal formulation loaded with tretinoin	Hydration media	% encapsulation efficiency (SD)
1	Span 60:CH (69:35) weight (mg) ratio	Water	24.5%±0.65%
2	Span 60:CH (69:35) weight (mg) ratio	Saline	43.25%±0.35%
3	Span 60:CH (138:35) weight (mg) ratio	Water	48.25%±0.82%
4	Span 60:CH (138:35) weight (mg) ratio	Saline	74.00%±0.72%
5	Span 60:CH (138:52) weight (mg) ratio	Water	52.05%±0.85%
6	Span 60:CH (138:52) weight (mg) ratio	Saline	86.45%±0.54%
7	Span 60:CH (207:52) weight (mg) ratio	Water	74.75%±0.34%
8	Span 60:CH (207:52) weight (mg) ratio	Saline	96.25%±0.56%
9	Span 60:CH (276:52) weight (mg) ratio	Water	61.25%±0.63%
10	Span 60:CH (276:52) weight (mg) ratio	Saline	75%±0.43%

Notes: Stock solution of tretinoin 4 mg/mL. Percent encapsulation efficiency is the mean from triplicate experiments. **Abbreviations:** CH, cholesterol; SD, standard deviation.

Stability studies of the niosomal formulation

The ability of vesicles to retain the drug was assessed by keeping the niosomal gel at three different temperature conditions, ie, refrigeration temperature ($4^{\circ}C-8^{\circ}C$), room temperature ($25^{\circ}C\pm2^{\circ}C$), and oven temperature ($45^{\circ}C\pm2^{\circ}C$). Throughout the study, niosomal gel formulations were stored in aluminum-foil-sealed glass vials. The samples were withdrawn at different time intervals over a period of 1 month and drug leakage from the formulations was analyzed for drug content by using a UV spectrophotometer.²⁶

In vitro permeation study

Permeation study of the prepared antiacne loson gel In vitro skin permeation studies were performed using Franz diffusion cells (Rama Scientific, Nov Del dia) with an effective diffusion area of 2.54 cm² e study was nducted using shaved Wistar rat skin. The skin was hounted on the receptor compartment with the strate of corneum site facing upwards into the donor compartme The dor or compartment was filled with 200 mg of the antic ne ni omal gel containing 0.020% 25-mL tretinoin and 0.600° BPO. guot of 1:1 (ethanol/ ptor medium to maintain methanol:salip v/v wa used as tor compartment was maintained at a sink cond. In. The a magnetic bar at 600 rpm. At appropriate 37°C and stirre liquots of the receptor medium were withtime intervals, 3-m. drawn and immediately replaced by an equal volume of fresh receptor solution up to 24 hours. The samples were analyzed by a UV spectrophotometer at 234.8 nm for BPO and 348.6 nm for tretinoin. The flux was calculated for each component from the niosomal gel formulation using Wistar rat skin.

Permeation study of the prepared antiacne cream

The donor compartment was filled with 200 mg of antiacne cream containing 0.020% tretinoin and 0.600% BPO. A 25-mL aliquot of 1:1 (ethanol/methanol:saline) v/v was

In to mai tain a s used as the receptor medi ondition. The receptor compartment my stained at 37°C and stirred a 600 rp. At appropriate time intervals, by a magnetic bar am were withdrawn and 3-mL aliquot receptor h y an equal volume of fresh receptor immediately replace. ap 524 hours. samples were analyzed by a UV soluti spe ophotometer at 234.8 nm for BPO and 348.6 nm for oin. The release rate flux was calculated for each comtret eam formulation using Wistar rat skin. from the ponen

Percention study of the prepared alcoholic solution ine donor compartment was filled with 200 µL of the antiacne lcoholic solution containing 0.020% tretinoin and 0.600% aPO. A 25-mL aliquot of 1:1 (ethanol/methanol:saline) v/v was used as the receptor medium to maintain a sink condition. The receptor compartment was maintained at 37°C and stirred by a magnetic bar at 600 rpm. At appropriate intervals, 3-mL aliquots of the receptor medium were withdrawn and immediately replaced by an equal volume of fresh receptor solution for up to 24 hours. The samples were analyzed by a UV spectrophotometer at 234.8 nm for BPO and 348.6 nm for tretinoin. The flux was calculated for each component from the niosomal gel formulation using Wistar rat skin.

In vitro skin-retention study

The ability of vesicles to help retain the drug within the skin milieu (ie, the depot effect) was investigated by determining the amount of drug retained in the skin samples employed in permeation studies. After completion of the permeation experiment, skin mounted on the diffusion cell was removed. The skin was cleaned with cotton, dipped in saline solution, and blotted with tissue paper to remove any adhering formulation. Subsequently, the skin sample was homogenized with 20 mL of a chloroform:methanol/ethanol mixture (2:1, v/v), for the extraction of a homogenate suspension which was thus obtained

using filter paper. For determining the amount of drugs retained in skin milieu, skin was subjected to homogenization using chloroform:methanol/ethanol mixture (2:1, v/v) for extraction of retained drugs, extracted drugs in chloroform:methanol/ethanol mixture (2:1, v/v) after homogenization was filtered using filter paper. The drug content was quantified using a UV spectrophotometer at respective absorption maxima for BPO and TRA.²⁷

In vivo study

The rabbit ear model was used to study comedone formation in order to assess the comedogenicity of cosmetics, toiletries, and dermatological preparation and to evaluate the potential of antiacne drugs. This comedo induction took place after about 2 weeks of repeated topical application of a chemical comedone such as 50% oleic acid. One set of rabbits was treated as a control and received no treatment, while the remaining two set of rabbits received treatment with 50% oleic acid and dimethyl sulfoxide for up to 28 days on the ventral aspect of the pinnas once a day. The total number of animals used was nine. A group of three animals were used in each of the three groups; one group was named as the control group, and the two other groups were used in the study. One of the test groups was treated with a niosomal gel and the other test groups was treated with cream. Dur study, all animals were subjected to histological examin ion for assuring the effectiveness of the tested for lation.

Stability of final niosomal g

Testing of the stability of the niosonal generated performed in triplicate. The niosomal suspenses was prepared separately for both the drugs and was dispersed to the gel. Three batches were prepared and studied at three temperature conditions (room temperature, refrigerated temperature, and 45°C) to evaluate the impact of storage temperature on the stability of niosomes dispersed in the gel.

Stability studies were performed by considering a worst-case condition for the formulation in terms of the maximum interaction of excipient by minimizing the concentration of drug-loaded niosomes. In the performed study, a lower quantity of drug-loaded niosomes were dispersed in the gel and kept under the mentioned storage temperature condition to evaluate the combined effect of temperature and excipient over the picture gel.

The stability of the final for alation w determined by placing the formulation in a stand condition a oom temperature, refrigerated temper are, and 5°C for 1 honth, and the content of the added rug in the form on was measured at various time intends () 5, 7, 14, and 30 days); changes in the content the fine ormulation were determined using hotometer (a UV spe nic 2201).

Recults and liscussion icompatibility studies between drugs

was clear from infrared spectroscopy that the mixture of TRA and 3PO was compatible because there were no alteration in band spectra when compared with each separate of the drug. Two sharp bands were observed for TRA (1,685.87 cm⁻¹ for C=O str and 2,937.68 cm⁻¹ for O-H str) and two bands for BPO (1,759.4 cm⁻¹ C=O str ester, 1,226.7 cm⁻¹, C-O str) (these are infrared spectroscopy band representative for functional group present in drugs. In TRA C=O and O-H group and BPO C=O for ester group and C-O is present) in the mixture of TRA and BPO (Figure 1).

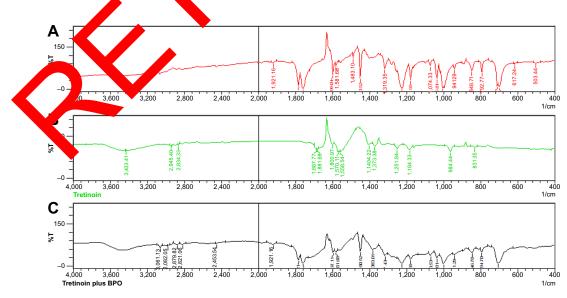


Figure I IR spectra of BPO, tretinoin, and the mixture of BPO and tretinoin. Notes: (A) BPO; (B) tretinoin; (C) mixture of BPO and tretinoin. Abbreviations: BPO, benzoyl peroxide; IR, infrared spectroscopy; T, transmittance.

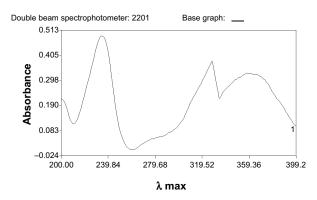


Figure 2 UV scan of the BPO and tretinoin mixture.

Abbreviations: BPO, benzoyl peroxide; max, maximum; UV, ultraviolet.

UV scan analysis also confirmed that the mixture of TRA and BPO were compatible because, upon scanning, the mixture of TRA and BPO ($10 \mu g/mL$) in the range of 200- to 400-nm TRA gave absorption maxima at 348.6 nm and 234.8 nm, respectively. The characteristic peak for TRA was obtained at 348.6 nm and for BPO it was obtained at 234.8 nm. The interference of absorbance of one component with the absorbance of tretinoin at 234.8 nm was -0.059 and for BPO at 348.6 nm it was -0.026. The results showed that there was no interference in the absorbance of each component in the mixture (Figure 2).

Susceptibility testing of BPO against S. epidermidis

In this study, BPO was examined for antibacterial activity against *S. epidermidis*. The susceptible estudy of the (Table 3; Figure 3) showed that BPO could effect only inhibit the growth of *S. epidermidis*.

Minimum inhibitory concentration of BPO for S. epidermidin

The evaluated data deconstructed that BPO is effective against S. epiderarchis at $^{\circ}$ µg/m², which shows that

28 μ g is the minimum concentration of BPO that will be effective in killing *S. epidermidis* at the affected/pathogenic site (Figure 4).

Characterization of niosomes

Morphology analysis by transmission electron microscopy

Transmission electron microscopy (TEM) was performed to determine vesicle formation and morphology. It was clear from the TEM analysis that uniform spherical niosomes were formed (Figures 5 and 6) and niosome photographs were taken from a unotomic escope at a $100 \times$ magnification (Figures \times and 8) confirming the same. TEM micrographs moving the average size of niosomes without drugatemonstrated user in the range of 200–250 nm.

Particle size massis by phone recorrelation spectroscopy (DL

DLLS ses showed that some of the samples were spersed (polydispersivity index =0.60) and the repropoly ility of vestme sizes appeared to be good. The mean duc diam r of trety in-saturated and BPO-saturated vesicles DLLS analysis, drug-containing noisomes, vas 616 m in gel then it was analyzed for estimating the size dig stribution through out the gel, which was showing that 616 m average diameter of noisome present in the niosomal el (Figure 9).

Zeta potential

The niosomal formulation containing drug-loaded niosomes (which was subjected to zeta potential analysis) had a zeta potential value of -43 mV, which is a measure of net charge of the niosomes (Figure 10). This higher charge on the surface of vesicles produced a repulsive force between the vesicles which

Serial number	Concentration µg/mL	Average of zone of inhibition (mm)	Standard deviation
1	50	12.5	±0.057
2	100	12.5	±0.057
3	150	13.9	±0.1
4	200	14.9	±0.115
5	250	14.8	±0.152
6	300	14.9	±0.1
7	350	14.9	±0.1
8	400	15.5	±0.1
9	450	15.5	±0.2
10	500	17.0	±0.057
11	1,000	16.9	±0.152
12	1,500	17.0	±0.152
13	2,000	16.9	±0. I

Table 3 Zone

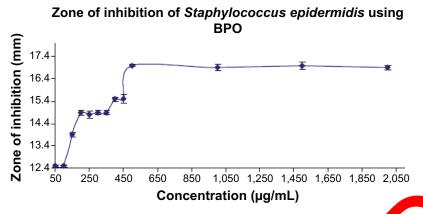


Figure 3 Susceptibility testing graph: BPO concentration versus zone of inhibition. Abbreviation: BPO, benzoyl peroxide.

made them stable and devoid of agglomeration and faster settling, providing an evenly distributed suspension. From this, it can be concluded that the present niosomal formulations show good stability. In the present study, both drugs were loaded in the niosomes vesicles separately and then further dispersed in the gel formulation for application onto the skin. Drug loading will have an impact on zeta potential because the zeta potential indicates the degree of repulsion between adjacent, similarly charged particles in dispersion. Due to this consideration, separate loading of drugs in niosomes was complete processed subsequently for gel preparation.

Encapsulation efficiency

Under the same preparation condition the er efficiency for both drugs was calculated. ncapsulation efficiency of BPO and treting s given in bles 1 and 2. Saline media for hydrax on way hosen for increasing the tretinoin encapsulation efficiency cause the desired required in niosomal gel formulation concentration of TP is less in comparise to B². The addition of a hypertonic suspection of rosomes brings about a salt solution reduction er of n. es, which leads to increased 1 diam

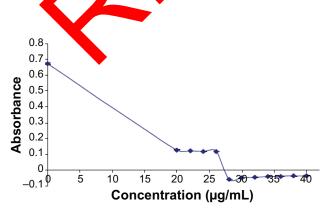
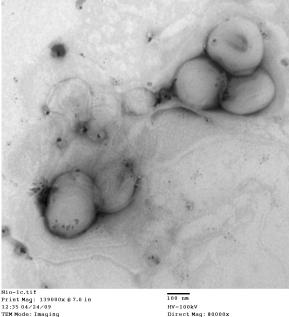


Figure 4 Minimum inhibitory concentration of benzoyl peroxide against Staphylococcus epidermidis.

h propertiona surface area and w ads to increased encapsulation effici cy.

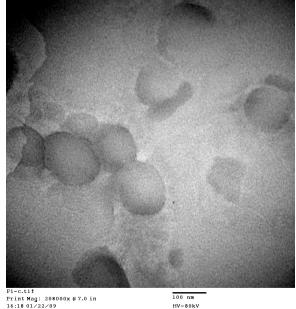
DSC anal s. for the de mation of the gel–lipid transition tem, rature

casurement were performed to determine the DS el-lipid transition temperature or phase transformation f niosomes. With the increase of system temperature, hydrocation chains in the ordered bilayer of vesicles normed from a rigid gel to facile liquid crystals were . ning that the phase transition of the bilayer took place, indicating that the niosome transitioned to a liquid crystal. The transition of the niosomal formulation from a gel state (ordered state) to a liquid crystalline state (disordered state)



Direct Mag: 80000x SAIF Punjab University Chandigarh

Figure 5 Transmission electron microscopy photograph of niosomes (negative staining).



TEM Mode: Imaging

HV = 80 kV Direct Mag: 120000x SAIF Punjab University Chandigarh

Figure 6 Transmission electron microscopy photograph of niosomes.

occurred at 56°C–58°C (Figure 11); this temperature range is, thus, the gel-lipid transition temperature of niosomes formed from span 60:cholesterol these ratio of Span 60: cholesterol (207:52 and 138:52 mg) used for preparation of TRA a BPO drug loaded niosomes respectively.

Stability studies of the niosomal formulation JN . Results showed that the niosomal gel form. It ion w stable at refrigeration and room temperatures e limited leakage of the drug was found at e temperal s. The drug retained at 45°C might have decreased due to the melting of the surfactant and lipid resent in the Norulation (Fig-, the posomal gel formulations ures 12 and 13). Theref can be stored at either riger ion or room temperature.

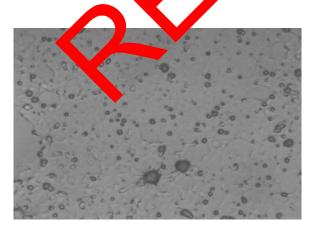


Figure 7 Photograph of benzoyl peroxide drug niosomes using a photomicroscope: 100× magnification.

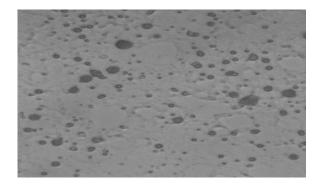


Figure 8 Photograph of tretinoin drug niosomes using a photomicroscope: 100× magnification.

In vitro permeation study In vitro permeation stude of nioso al gel

The mean amount of testinoin ad BPC ermeated per unit of surface area s determined during 24-hour experiments. Figure 14 shows the remeation profiles of the niosomal a mulative a Ints of tretinoin and BPO permeated versus tin. The permeated amount of tretinoin from te mosomal gel ther 24 hours was calculated as 6.2 $40.14 \,\mu\text{g/cm}^2$ and $5.04 \pm 0.014 \,\mu\text{g/cm}^2$ was calculated as permeated mount of BPO from the niosomal gel after 2 hours.

Pe tion study of the prepared antiacne cream gure 14 also shows the permeation profiles (the cumulaive amounts of tretinoin and BPO permeated versus time) rough the skin obtained from tretinoin and BPO containing o/w cream. As can be seen in Figure 12, the permeation curves do not show a classic profile with a steady-state phase. The maximum permeated amount of tretinoin from the cream after 8 hours was calculated as $6.60\pm0.13 \,\mu\text{g/cm}^2$, and 7.91 \pm 0.023 µg/cm² was the calculated permeated amount of BPO from the cream after 24 hours. Eight hours for tretinoin and 24 hours for BPO was considered the time taken for maximum amount permeated through skin using cream formulation. Cream is the w/o system; containing a major proportion of oil phase and due to the lipophilic nature of skin, the permeation content from cream is faster, which is the reason for faster permeation of tretinoin within 8 hours.

Permeation study of the prepared alcoholic solution Figure 14 also shows the permeation profiles (the cumulative amounts of tretinoin and BPO permeated versus time) through the skin obtained from tretinoin and a BPO alcoholic solution. The maximum permeated amount of tretinoin from the alcoholic solution after 6 hours was calculated as 7.72 \pm 0.16 µg/cm², and 12.18 \pm 0.013 µg/cm² was the

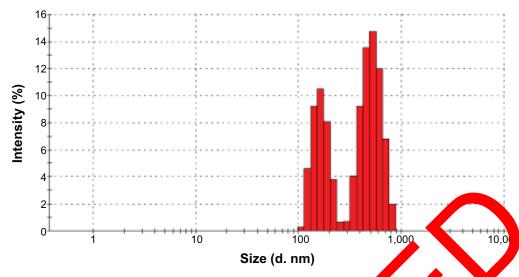


Figure 9 Statistical bar graph of particle size distribution in niosomal suspensions.

calculated permeated amount of BPO from the alcoholic solution after 6 hours.

In vitro skin-retention study

Results also showed that the drug content retained in the layers of skin was 11.54 µg of tretinoin and 68.85 µg of BPO from the cream, 2.68 µg of tretinoin and 59.98 rg of BPO from the alcoholic solution, and 15.54 μ g of tre 10111 and 143.78 µg of BPO from the niosomal gel. The ug he cre retention was more in case of niosomal get allah Juded h and alcoholic solution, so it can be con e for th first time that the gel was more effe tive an alcoholic solution (Figure 15).

In vivo studies

the histological studies showed Results obtained fre that the prepared osomal formulation was effective in the treatment of ach mparise of the control sample 'hat ' ere was a marked increase and treated Allh howe e of the sebaceous gland and several units of in volv comedo sent man treated pinna. Moreover, for the n acne-induce pinna treated with the prepared niosomal gel

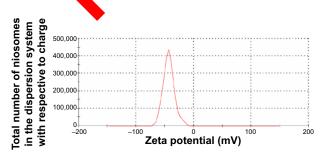


Figure 10 Zeta potential of niosomal suspensions.

ays, histo. eports showed that there for up to 1 rica1 duction in the volume of the sebaceous rkea was a r gland and no dila. ion was present in the follicle (Figure . Similar results were obtained with the cream which of BPO and 0.025% of tretinoin. From this ontains 2.5 tudy, it can be concluded that the niosomal mparative e efficacious than the cream. It showed a simige r therapeutic action when using a 4.16-fold lower dose of **D**. Moreover, histopathology micrograph showed that no comedones were present in the treated pinnas after treatment for 14 days with both antiacne cream and the niosomal formulation.

Conclusion

In summary, it is evident from the aforementioned study that niosomes showed better therapeutic activity than conventional dosage forms using formulations through the same route of administration. The greatest challenge with topical drug delivery is the barrier nature of skin, which restricts

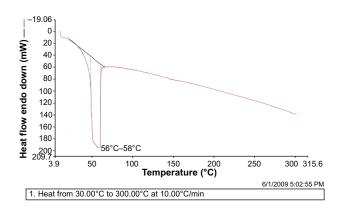
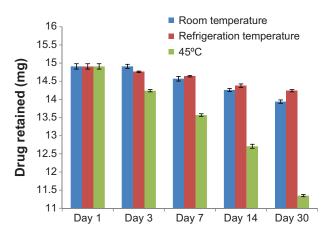
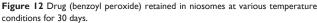


Figure 11 Differential scanning calorimetry thermogram of the niosomal preparation.





the entry of most drugs. Here, the present data proved that niosomes acted as the best vesicles in dermal drug delivery due to its nanometer size and their elastic nature. They acted as a drug carrier to deliver entrapped drug molecules into or across the skin and, owing to the individual lipid components, enhanced penetration into the stratum corneum and, subsequently, the alteration of the intercellular lipid lamellae within this skin layer. In vivo experiments demonstrated an interesting correlation between the better permeati capabilities of niosomes in comparison to other conventional dosage forms in terms of a better therapeutic efficacy at the affected site at lower doses of drugs present in e 111 mal gel formulation. Comparative in vivo studie of the night omal gel and antiacne cream of TRA and B O sh that the in the antiniosomal gel was more efficacious e cream



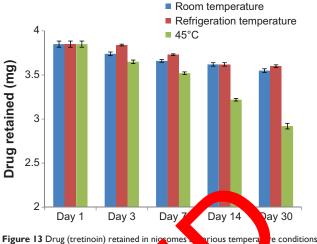


Figure 13 Drug (tretinoin) retained in nicromes a parious temperature conditions for 30 days.

because niosomal gele acrease the therapeutic index of reduction in dose of BPO in a drug leading to 1.16-A comparison to t ontiacne cre microbial susceptibilved that **B** has potent antibacterial ity and MIC Lata sh action against acne-cau g bacterium such as S. epidermidis at 28 g/mL. An ex vivo skin-retention study showed that the nio mal gel had aximum skin retention of BPO and TRA affected sit Due to maximum retention at the skin, at th derium will not propagate, and the niosomal acne-ca n maintain the MIC at the target site for prolonged 20 riods of time due to a niosomal "depot mechanism". Based on the above data, it can be concluded that the nano sicle (ie, niosomes)-based dosage forms developed here would have a better therapeutic efficacy at a lower dose in comparison to conventional dosage forms.

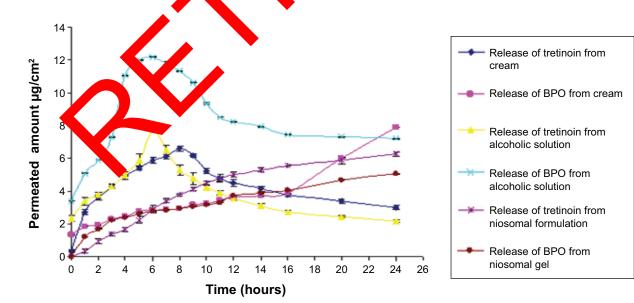


Figure 14 Permeation profile of BPO and tretinoin from cream, alcoholic solution, and niosomal gel. Abbreviation: BPO, benzoyl peroxide.

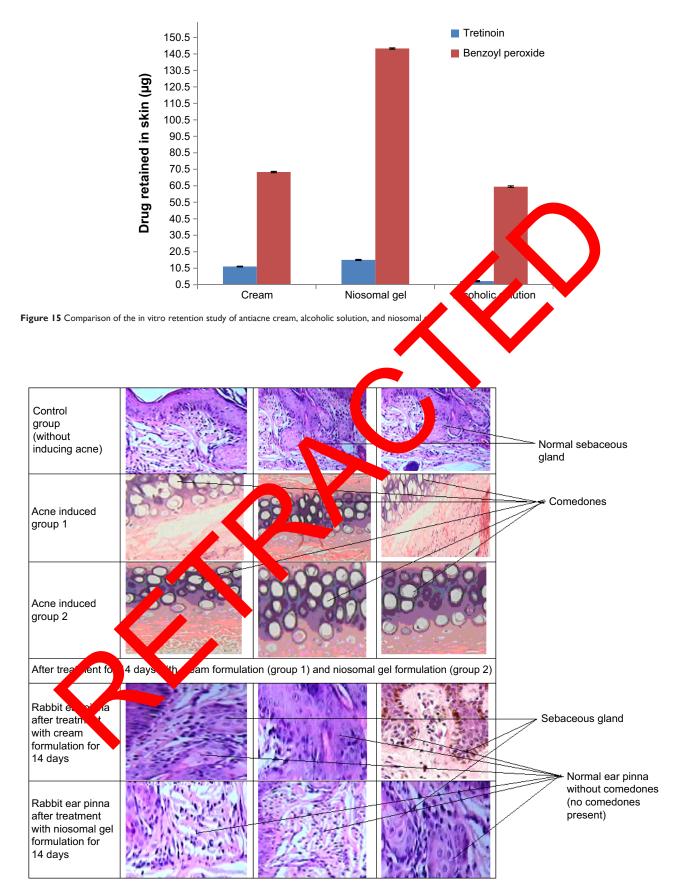


Figure 16 Comparative histopathological examination of control versus acne-induced pinna after treatment with cream for 14 days and after treatment with niosomal gel.

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Disclosure

The authors report no conflicts of interest in this work.

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