

Sunroot mediated synthesis and characterization of silver nanoparticles and evaluation of its antibacterial and rat splenocyte cytotoxic effects

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Abstract: A rapid, green phytosynthesis of silver nanoparticles (AgNPs) using the aqueous extract of *Helianthus tuberosus* (sunroot tuber) was reported in this study. The morphology of the AgNPs was determined by transmission electron microscopy (TEM). Scanning electron microscopy–energy-dispersive spectroscopy (SEM–EDS) and X-ray powder diffraction (XRD) analysis confirmed the presence of AgNPs. Fourier transform infrared spectroscopy (FTIR) analysis revealed that biomolecules in the tuber extract were involved in the reduction and capping of AgNPs. The energy-dispersive spectroscopy (EDS) analysis of the AgNPs, using an energy range of 2–4 keV, confirmed the presence of elemental silver without any contamination. Further, the synthesized AgNPs were evaluated against phytopathogens such as *Ralstonia solanacearum* and *Xanthomonas axonopodis*. The AgNPs (1–4 mM) extensively reduced the growth rate of the phytopathogens. In addition, the cytotoxic effect of the synthesized AgNPs was analyzed using rat splenocytes. The cell viability was decreased according to the increasing concentration of AgNPs and 67% of cell death was observed at 100 µg/mL.

Keywords: cytotoxicity, *Helianthus tuberosus*, nanobiotechnology, phytosynthesis, splenocytes

Introduction

Nanobiotechnology is a promising interdisciplinary field of biotechnology and nanoscience that offers novel nanoscale materials with interesting applications in different scientific fields such as medicine, biotechnology, chemistry, physics, and material sciences.^{1,2} Metallic nanoparticles are attractive due to their potential applications in novel technologies.³ Among the metallic nanoparticles, silver nanoparticles (AgNPs) have attracted the attention of researchers in the field of science and technology due to their vast biological applications. Several studies have reported the synthesis of AgNPs by physicochemical methods.^{4,5} The chemical synthesis of AgNPs may lead to the presence of some toxic chemicals on the surface that may have adverse effects in its application. Also, the chemical used in the synthesis may pollute the environment.⁶ Hence, there is an urgent need to develop eco-friendly biological methods of AgNPs synthesis instead of using toxic chemicals.

Nowadays, biological synthesis of AgNPs has been proposed as an emerging technology, and offers several advantages compared to conventional methods. Several studies reported AgNPs synthesis using biological materials, such as microorganisms, plant extracts, milk, and panchakavya.^{7–10} Phytosynthesis of AgNPs using plant materials has an edge over microbial mediated synthesis owing to their immediate and large-scale production.^{11,12} Thus, considerable attention is given to exploit the

synthesis of AgNPs using plant-based products. Several studies reported on the synthesis of AgNPs using various plant extracts.^{13–15} It has been established that proteins present in the plant extracts were involved in the reduction of Ag⁺ ions to nanocrystallites. *Helianthus tuberosus* L., a perennial herb, is a species of sunflower native to eastern North America and widely cultivated across the temperate zone for its edible tuber. Extracts of *H. tuberosus* L. tubers are aperient, cholagogue, and diuretic and have long been used in folk medicine to treat stomach problems, diabetes, and rheumatism.^{16,17} However, to our knowledge, the sunroot (*H. tuberosus* L.) tuber extract has never been used for the synthesis of AgNPs.

In vitro cytotoxicity study is an important assay to evaluate the mechanisms of toxicity caused by nanoparticles. AgNP-induced toxicity is related with mitochondrial damage, oxidative stress, DNA damage, and induction of apoptosis.¹⁸ Previous studies reported the cytotoxicity of AgNPs against NIH 3T3 fibroblast cells, HeLa cells, human glioblastoma cells, and human breast cancer cells (MCF-7).^{19–22} However, to our knowledge, cytotoxicity of AgNPs in rat splenocytes have never been explored.

Plant disease control is an important requirement for agriculture in the 21st century. Microorganisms are associated with several devastating diseases in economically important crops worldwide. Phytopathogenic bacteria cause enormous problems in agriculture, resulting in severe economic losses, since plants are the main nutrient sources of these pathogens.²³ *Ralstonia solanacearum* and *Xanthomonas axonopodis* are the most extensively studied phytopathogens in potato (*Solanum tuberosum*), tobacco (*Nicotiana tabacum*), and tomato (*Lycopersicon esculentum*) plant systems. The increasing population of these phytopathogens causes degradation of the occluded xylem vessels and the death of the plants. Hence, the synthesis of AgNPs and their antibacterial properties are emerging as fields of great interest among researchers. Hence, the objectives of the present study were (i) to synthesize AgNPs using *H. tuberosus* tuber extract, (ii) to characterize the synthesized AgNPs, and (iii) to assess the cytotoxicity of AgNPs synthesis against freshly isolated rat splenocytes, and (iv) to evaluate the bactericidal activities of the synthesized AgNPs.

Materials and methods

Plant material

The dried tuber of *H. tuberosus* was purchased from a local shop in Iksan, South Korea. One kilogram of tuber powder was soaked in 2.5 L methanol for 78 hours with occasional stirring. The solvent was removed by using Rotovac below

70°C. The solvent-free aqueous extract was used for the synthesis of AgNPs.

Synthesis of AgNPs

Silver nitrate (AgNO₃) was purchased from Sigma-Aldrich (St Louis, MO, USA) and the synthesis of AgNPs was carried out according to Lee et al.⁸ Briefly, 4 mL of the extract was mixed with 96 mL of 1 mM AgNO₃ solution and the resulting greenish white mixture was incubated for 8 hours in a rotary shaker (200 rpm) at 26°C. The reduction of Ag⁺ ions to Ag nanocrystals was monitored by the change in the color of the reaction mixture from greenish white to dark brown.

Characterization of AgNPs

The morphology of the synthesized AgNPs was examined using transmission electron microscopy (Bio-TEM) (H-7650; Hitachi Ltd., Tokyo, Japan). The elemental composition of the synthesized AgNPs was confirmed by scanning electron microscopy–energy-dispersive spectra (SEM–EDS) (JEOL-64000; Tokyo, Japan). The X-ray powder diffraction (XRD) was carried out using Rigaku X-ray diffractometer (Rigaku, Japan). The scanning was performed in the region of 2 θ =30°–80° at 0.041°/min with a time constant of 2 seconds. The Fourier transform infrared spectrum (FTIR) of the AgNPs was obtained on a PerkinElmer FTIR spectrophotometer (Waltham, MA, USA) in the diffuse reflectance mode at a resolution of 4 cm⁻¹ in KBr pellets.

Antibacterial activity of AgNPs

The phytopathogenic bacterial strains *R. solanacearum* and *X. axonopodis* were procured from the Korean Agriculture Culture Collection (KACC), South Korea. The freshly cultured bacterial strains from the Luria-Bertani (LB) agar plates were inoculated into LB broth and incubated at 37°C in a shaking incubator. After appropriate growth, the cultures were used for further experiments. The cultures were allowed to grow in 100 mL of LB broth containing the synthesized AgNPs at different concentrations in the range 1–4 mM. The optical density was measured every 4 hours to determine the growth of the bacteria using the Shimadzu UV-1800 spectrophotometer. The culture without AgNPs was used as a control.

Isolation and propagation of rat splenocytes

Adult (Sprague dawley, 8–12 week old) rats were purchased from Koatech, South Korea. The rats were maintained in a specific pathogen-free facility. Fresh splenocytes of the rat was obtained by teasing the spleen under aseptic

conditions according to Lu et al.²⁴ Single-cell suspensions were prepared from rat spleen by pressing the tissues through a sterile wire mesh and washing the cells in Roswell Park Memorial Institute (RPMI) medium containing 1% antibiotic (Anti–Anti; Gibco, South Korea). The animal experiments were carried out in accordance with the institutional animal care and use committee at Chonbuk National University.

3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide assay

The isolated rat splenocytes were treated with series of 10–100 µg/mL of phytosynthesized AgNPs in 96-well tissue culture plates. The treated cells were incubated for 24 hours at 37°C with 5% CO₂ for cytotoxicity analysis. The cells were then subjected to 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay. The stock concentration (5 mg/mL) of MTT, a yellow tetrazole, was prepared and 20 µL of MTT was added in each AgNP-treated well and incubated for 4 hours. Purple colored formazan crystals were observed in the bottom of the well and these crystals were dissolved with 200 µL of dimethyl sulfoxide (DMSO), and read at 595 nm^{18,22} in a multi-well plate reader (Epoch microplate spectrophotometer; Biotek, Winooski, USA).

Results and discussion

Characterization studies

The addition of the tuber extract to 1 mM-solution of AgNO₃ changed the color from greenish white to dark brown in about 1 hour. The intensity of the color increased after 8 hours of incubation and the reduction of pure Ag⁺ ions to Ag⁰ was monitored by measuring the UV–Vis spectrum of the reaction media (Figure 1). The UV–Vis spectra of the silver surface plasmon resonance band occurs near 430 nm. Previous studies

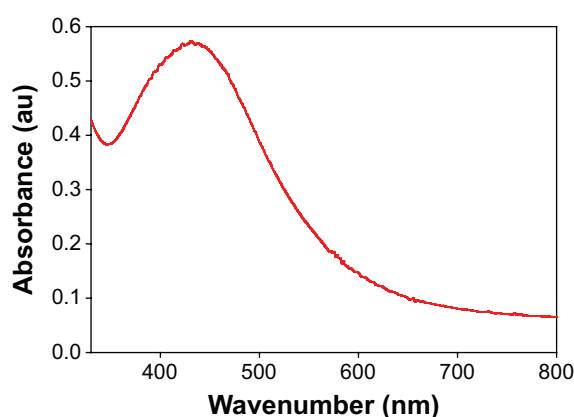


Figure 1 UV–Vis absorption spectrum of the AgNPs prepared from 1 mM AgNO₃ solution.

Abbreviation: AgNPs, silver nanoparticles.

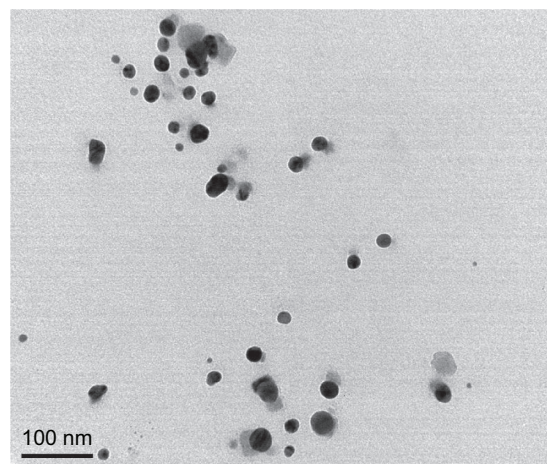


Figure 2 TEM image of AgNPs.

Abbreviations: TEM, transmission electron microscopy; AgNPs, silver nanoparticles.

have reported the presence of AgNPs exhibiting yellowish brown color in solution due to the excitation of surface Plasmon vibrations.^{25,26} However, the color change was not observed in the control flask. The biomolecules present in the tuber extract may reduce the Ag ions present in the solution. This was supported by the results from FTIR studies where stretching vibrations of amines, alkanoids, and alkaloids were observed.

The morphology and size of the phytosynthesized AgNPs were observed by TEM images (Figure 2). The particles were spherical in shape, monodisperse, and the size of the particles varied from 10–70 nm. To confirm the presence of Ag, the samples were analyzed in SEM–EDS, and the results are shown in Figure 3. The results showed strong silver signals (3 keV), along with weak oxygen and carbon peaks, which might have originated from the tuber extract. The results are

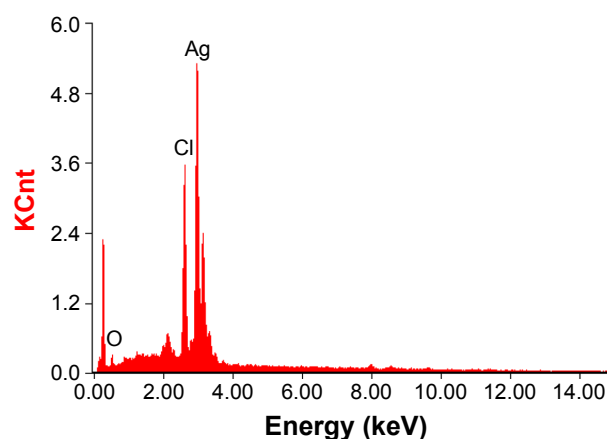


Figure 3 SEM–EDS spectrum of AgNPs. A strong peak at 3 keV confirms the presence of Ag.

Abbreviations: SEM–EDS, scanning electron microscopy–energy-dispersive spectroscopy; AgNPs, silver nanoparticles.

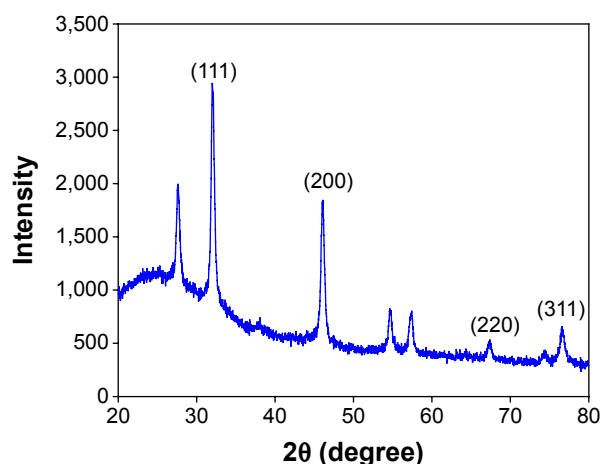


Figure 4 XRD pattern of AgNPs synthesized using tuber extract.
Abbreviations: XRD, X-ray powder diffraction; AgNPs, silver nanoparticles.

consistent with previous studies that reported the strong peak for AgNPs at 3 keV.^{7,8} The EDS quantitative analysis showed the presence of silver (100%) without any contaminants.

XRD was used for the identification of the crystal nature of the AgNPs. The XRD patterns of the synthesized AgNPs are shown in Figure 4. The peaks at 2θ values of 38, 46, 67.4, and 78 correspond to 111, 200, 220, and 311, indicating the cubic nature of AgNPs. The results are in agreement with several studies that reported the cubic nature of biologically synthesized AgNPs.^{27,28} The FTIR spectrum of the AgNPs is shown in Figure 5. Some pronounced peaks were observed at 3,329, 2,941, 1,759, 1,067, and 731 cm^{-1} in the 4,000–400 cm^{-1} region. The corresponding peaks were associated with the stretching vibrations of $-\text{C}-\text{O}$, $\text{C}-\text{H}$, $\text{C}=\text{C}$, CH_2 , and $\text{O}-\text{H}$, respectively. The absorbance

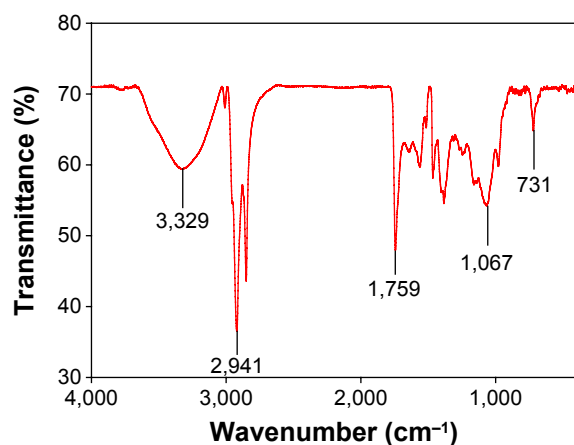


Figure 5 FTIR spectra of AgNPs synthesized using tuber extract.
Abbreviations: FTIR, Fourier transform infrared spectroscopy; AgNPs, silver nanoparticles.

peaks could be attributed to the phytochemicals present in the tuber extract, such as reducing sugars, flavonoids, saccharides, and proteins.²⁹ Pan et al¹⁶ reported the presence of coumarins, unsaturated fatty acids, and phenols in the *H. tuberosus* extract.

Antibacterial activity

Numerous studies reported that biologically synthesized AgNPs have significant antibacterial and antifungal activities, and could be used for the treatment of bacterial and fungal diseases in biotic communities.^{7,8} However, the adverse effects of AgNPs need to be carefully evaluated before they could be used in antimicrobial products. Hence, phytosynthesized AgNPs were studied for antibacterial activity against phytopathogenic bacteria, namely, *R. solanacearum* and *X. axonopodis*. The bacterial growth measurements at different concentrations (1–4 mM) of the AgNPs were determined 0–24 hours at regular time intervals (Figure 6A and B). The observed results indicated that 4 mM concentrations of the AgNPs effectively encountered the bacterial population

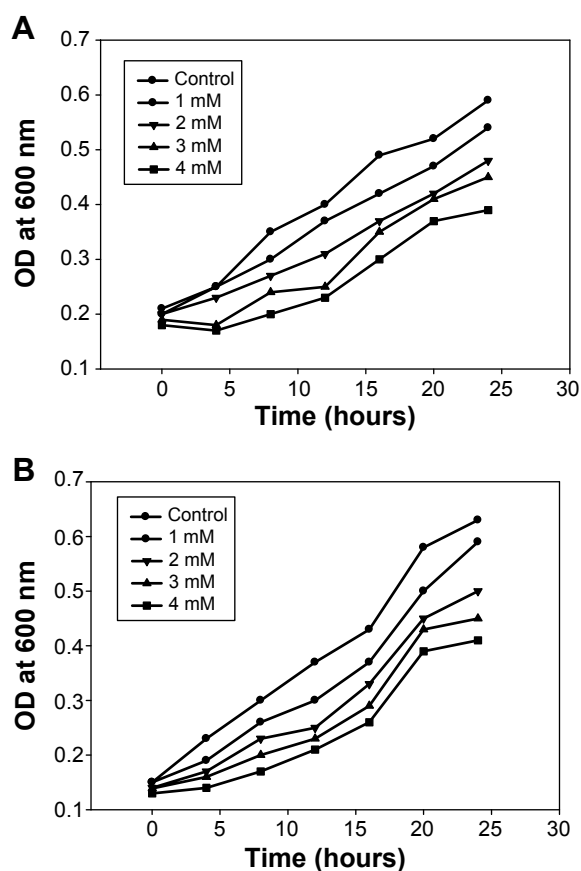


Figure 6 Effect of AgNPs on growth of phytopathogenic bacteria.
Notes: (A) *Ralstonia solanacearum* and (B) *Xanthomonas axonopodis*.
Abbreviation: AgNPs, silver nanoparticles.

in the medium. The results are consistent with our previous study reporting the antibacterial activity of AgNPs against antibiotic resistant strains.⁷

Cytotoxicity

The cytotoxic potential of the biologically synthesized AgNPs was assessed using MTT assay. Figure 7A shows the viability of splenocytes with the increasing concentration

(10–100 $\mu\text{g/mL}$) of AgNPs, and the reduction of color compared to untreated cells observed at 595 nm revealed the cytotoxicity. Thus, in vitro cytotoxicity of the AgNPs was evaluated against rat splenocytes at different concentrations (10, 20, 40, 80 and 100 $\mu\text{g/mL}$), and the results are shown in Figure 7B. The results clearly demonstrated that the splenocytes viability was directly proportional to the concentration of the AgNPs. The cell death (10%) was

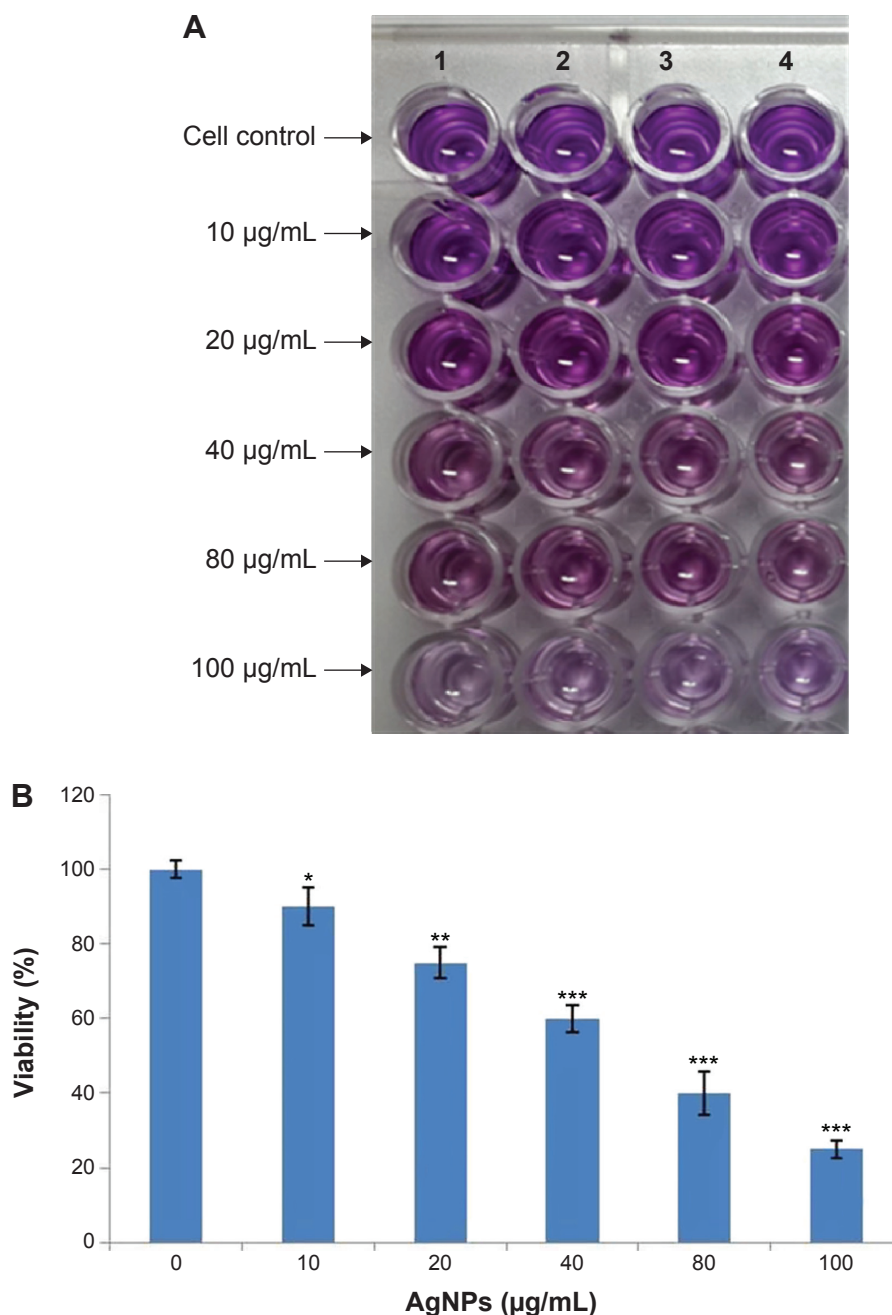


Figure 7 Cytotoxic effect of AgNPs on growth of rat splenocytes.

Notes: (A) a 96-well plate image of MTT assay, the toxicity is indicated by color reduction; (B) dose dependent reduction of cell viability at 595 nm. * $P < 0.05$; ** $P < 0.01$; *** $P < 0.001$.

Abbreviation: AgNPs, silver nanoparticles.

reported at 10 µg/mL and gradually increased according to the concentration and reached the maximum (67%) at 100 µg/mL. Hackenberg et al³⁰ reported that 10 µg/mL of AgNPs reduced human mesenchymal stem cells viability within 1 hour exposure. Vivek et al²² reported that the cytotoxicity of AgNPs was significantly increased according to the concentration of nanoparticles. Moreover, several studies reported that AgNPs may induce reactive oxygen species and cause damage to cellular components leading to cell death.^{31–33}

Conclusion

To the best of our knowledge, this is the first study to report the synthesis of AgNPs using *H. tuberosus* tuber extract. The proposed method is a simple, green, and cost-effective method for the synthesis of AgNPs without any harmful chemicals. The analytical results confirmed the presence of AgNPs without any contamination. The green synthesized AgNPs had a significant bactericidal property against phytopathogenic bacteria. The cytotoxicity of the synthesized AgNPs indicates that further studies are needed before using these nanoparticles in antimicrobial products.

Acknowledgment

This research was supported by Basic Science Research Program through the National Research Foundation of Korea (NRF) funded by the Ministry of Education, Science and Technology (2012R1A1A4A01011658).

Disclosure

The authors declare no conflict of interest in this work.

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