

Zein-alpha lipoic acid-loaded nanoparticles to enhance the oral bioavailability of dapoxetine: optimization and clinical pharmacokinetic evaluation

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Background: Premature ejaculation (PE) is the most common type of male sexual disorder with important psychological consequences. Dapoxetine (DPX), a recently approved drug for the treatment of PE, suffers from low bioavailability with large variability that ranges from 15–76% (mean 42%) after oral administration. The objective of this study is to optimize the parameters for the preparation of DPX-Zein-alpha lipoic acid (ALA) nanoparticles (NPs) to improve the bioavailability of DPX and consequently decrease therapeutic dose and adverse effect, leading to patient satisfaction and compliance.

Methods: We investigated the effect of ALA concentration, PVA concentration and stirring rate on nanoparticle size (Y_1), zeta potential (Y_2), initial DPX release (Y_3) and cumulative DPX release (Y_4). In addition, in vivo pharmacokinetic study was performed for the optimized DPX formulation on human healthy volunteers compared with marketed DPX tablet.

Results: The optimized DPX-loaded NPs showed Y_1 , Y_2 , Y_3 , and Y_4 of 159.24 nm, 19.14 mV, 25.31% and 95.9 %, respectively. A single oral dose of 30 mg of optimized DPX-loaded NPs to human volunteers resulted in 2-fold improvement of AUC (1376.145 ± 339.592 vs 709.178 ± 146.307 in DPX), 4-fold increase in t_{max} (2.5 ± 0.314 vs 0.583 ± 0.144), prolongation of MRT (7.637 ± 1.373 compared to 6.031 ± 1.826 h), but with reduction in $t_{1/2}$ (5.283 ± 1.077 vs 8.452 ± 2.813).

Conclusion: The clinical findings suggest 194% enhancement of relative bioavailability of the optimized DPX-loaded NPs, potentially leading to a decrease in therapeutic dose and associated side effects in the treatment of PE.

Keywords: dapoxetine, premature ejaculation, mathematical experimental design, nanoparticles, clinical pharmacokinetics, bioavailability

Introduction

The Global Study of Sexual Attitudes and Behaviors (GSSAB) has reported the Premature ejaculation (PE) to be approximately 30% across all age groups.^{1–3} PE is associated with serotonin neurotransmission imbalances, which is usually not related to other chronic disorders.⁴

Dapoxetine (DPX), with an antidepressant property, is the first serotonin transporter inhibitor successfully used for the treatment of PE.^{5–7} DPX increases the action of serotonin at the postsynaptic cleft. The fast-acting effect and rapid elimination of DPX make it a good candidate for the treatment of PE, but not as an antidepressant. The efficacy and safety of DPX in the treatment of PE were confirmed in 2011 and granted marketing

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authorization for 30 mg and 60 mg film-coated tablets in the European Union.⁸ Nonetheless, DPX suffers from low bioavailability with large variability 15–76% (mean 42%) after oral administration. DPX is extensively metabolized through several pathways, for example Liver and kidney to give about 40 metabolites that are essentially inactive. Accordingly, formulation of DPX as Zein nanoparticles prolongs its circulation in the body and protects DPX from metabolism, hence improve DPX bioavailability and efficacy.⁹ The peak plasma concentration of DPX after oral administration is reached within 60–80 min and is dose-related.^{10,11} Nausea, dizziness, headache, insomnia, fatigue, and nasopharyngitis represent the most commonly treatment-related adverse effects of DPX,¹² which were also dose-dependent and mostly synchronized with the pharmacokinetic profile of DPX.⁷

Zein is used as a matrix for nano/and micro-structured drug delivery systems.^{13–16} Zein contains high percentage of hydrophobic amino acids that renders Zein insoluble under physiological conditions that cause aggregation into colloidal particles. The hydrophobic residues of Zein polymer (as lucine, phenylalanine, etc.) allows the self-association of Zein protein into bodies that are stably retained in membrane vesicles.^{13–16} The hydrophobic nature of Zein and excellent biocompatibility allows its use to design and sustain the release of loaded drugs.^{17,18}

Alpha-lipoic acid (ALA) is a powerful antioxidant that is also known as acetate replacing factor. ALA improves sexual function and used in the treatment of diabetic impotence due to its antioxidant activity.^{19–23} Recently, ALA has been used as a formulation excipient to improve the delivery and efficacy of drugs.^{24,25}

The aim of this study was to use Zein and ALA to encapsulate DPX and utilize mathematical experimental design to optimize formulation parameters for the preparation of DPX-Zein-ALA NPs. The objective is to control DPX release utilizing Zein as a retarding polymer which increasing its residence time in the body and hence improving bioavailability of DPX after oral administration to decrease its therapeutic dose and hence decrease the incidence of its adverse effects.

Materials and methods

Materials

DPX was a gift from Spimaco Addwaeih (Riyadh, KSA). ALA, Zein, polyvinyl alcohol (PVA), acetonitrile and methanol were purchased from Sigma-Aldrich Corporation (St. Louis, Missouri, USA).

Experimental design of DPX formulations

Based on preliminary data, the factors of ALA percentage in Zein (X_1), PVA percentage concentration (X_2) and stirring rate (X_3) were selected to study their effect on the particle size (Y_1), zeta potential (Y_2), initial DPX release (Y_3) and cumulative DPX release (Y_4) of the prepared formulations. The Box-Behnken design was suggested for the development of DPX formulations using Statgraphics Centurion XV version 15.2.05 software (StatPoint Technologies Inc., Warrenton, VA, USA). The goals for the investigated dependent variables of the experimental design were set to minimize Y_1 and maximize Y_2 , Y_3 , and Y_4 . The independent variables levels and the dependent variables constraints are shown in Table 1.

Table 1 Independent (X_1 – X_3) and dependent (Y_1 – Y_4) variables of DPX-Zein-ALA nanoparticles formulations utilized in Box-Behnken design

Independent Variables	Unit	Levels		
		–I	0	I
Percentage of ALA in Zein (X_1)	%	15	20	25
Percentage of PVA (X_2)	%	0.5	1	1.5
Stirring rate (X_3)	rpm	800	1000	1200
Dependent variables	Unit	Constraints		
		Low	High	Goal
Particle size (Y_1)	nm	148.6	206.8	Minimize
Zeta potential (Y_2)	mV	0.294	19.1	Maximize
Initial DPX release after 1h (Y_3)	%	18.09	35.67	Maximize
Cumulative DPX release after 24 h (Y_4)	%	79.05	97.65	Maximize

Abbreviations: DPX, dapoxetine; ALA, alpha lipoic acid; PVA, polyvinyl alcohol.

Preparation of DPX-loaded NPs

Fifteen different formulations of DPX-loaded NPs were prepared as previously reported.²⁴ Briefly, 30 mg of DPX, and the specified weight of Zein and ALA as described in Table 2 were stirred in 30 mL methanol at 1000 rpm at room temperature until completely solubilized. The methanolic solution was gradually added to 45 mL of an aqueous solution of PVA (0.5–1.5% w/v) and stirred for 4 hrs, then rotated in a rotary evaporator until complete evaporation of the methanol. The colloidal NPs dispersion was centrifuged at 20,000 rpm for 45 mins at 8 °C and washed with deionized water, then lyophilized using freeze dryer Alpha1-2/LDPlus (Martin Christ, Osterode am Harz, Germany) for 72 hrs. Mannitol (5 % w/v) was added to the colloidal dispersion as a cryoprotectant and finally stored until fully characterized.

DPX-loaded NPs characterization

Particle size and zeta potential analysis

Samples of the prepared NPs formulations were diluted with 20 times their volume and vortexed for 1 min and evaluated for particle size, size distribution and zeta potential using Malvern zetasizer nano zsp analyzer (Malvern

Panalytical, Enigma Business Park, Malvern, United Kingdom).

DPX encapsulation efficiency, DPX loading and yield percent

DPX encapsulation efficiency percent (EE %) was determined by dissolving accurately weighed amount of the NPs formulation in methanol, then sonicated and filtered. The encapsulated DPX in the formulation was measured by high-performance liquid chromatography (HPLC). EE %, DPX loading % and yield % were calculated according to Equations (1)–(3)

$$EE \% = \left(\frac{\text{Amount of drug in the formula}}{\text{Amount of drug initially added}} \right) \times 100 \quad (1)$$

$$DPX \text{ Loading \%} = \left(\frac{\text{Drug mass in nanoparticles}}{\text{mass of nanoparticles}} \right) \times 100 \quad (2)$$

$$Yield \% = \left(\frac{\text{mass of nanoparticles}}{\text{total mass of formula components}} \right) \times 100 \quad (3)$$

Table 2 Formulation variables (X_1 – X_3) of DPX-Zein-ALA nanoparticles and their observed responses (Y_1 – Y_4) as suggested by Box-Behnken design

Run #	Independent variables			Dependent variables*			
	Percentage of ALA in Zein (X_1)	Percentage of PVA (X_2)	Stirring rate (X_3)	Particle size (Y_1)	Zeta potential (Y_2)	Initial DPX release after 1h (Y_3)	Cumulative DPX release (Y_4)
DN-1	15	1.5	1000	155.6	17.3	29.68	97.32
DN-2	15	0.5	1000	159.7	19.1	31.34	97.65
DN-3	20	0.5	800	168.9	15.2	24.34	93.76
DN-4	25	1.0	800	205.7	0.41	19.1	87.61
DN-5	20	1.5	800	163.8	10.6	26.1	96.99
DN-6	25	0.5	1000	206.8	0.41	18.09	79.05
DN-7	20	0.5	1200	150.1	15.1	33.45	96.95
DN-8	15	1.0	1200	148.6	17.9	35.67	97.03
DN-9	25	1.5	1000	194.7	0.29	22.7	88.2
DN-10	25	1.0	1200	196.9	0.39	20.1	89.4
DN-11	20	1.5	1200	160.2	13.6	30.21	96.55
DN-12	15	1.0	800	167.3	18.0	23.54	93.21
DN-13	20	1.0	1000	179.1	14.6	28.48	89.54
DN-14	20	1.0	1000	177.4	14.3	29.36	91.11
DN-15	20	1.0	1000	176.9	14.7	27.12	92.89

Note: *The observed values of Y_1 , Y_2 , Y_3 and Y_4 represent the means of three determinations; standard deviations were <5% of the mean and thus were omitted from the table.

Abbreviations: DPX, dapoxetine; ALA, alpha lipoic acid; X_1 , percentage of alpha lipoic acid in Zein; X_2 , percentage of polyvinyl alcohol; X_3 , stirring rate; Y_1 , particle size (nm); Y_2 , zeta potential (mV); Y_3 , initial DPX release after 1h (%); Y_4 , cumulative DPX release after 24 h (%).

In vitro release of DPX-loaded NPs

In vitro release study was investigated utilizing Franz diffusion cell (automated diffusion system, Microette Plus, Hanson research, Chatsworth, CA, USA) as previously described.²⁶ Briefly, the specified weight of DPX-loaded NPs in the donor compartment above the diffusion membrane (0.1 μ m, PALL Corporation, NY, USA) was allowed to diffuse into phosphate-buffered saline pH 7.0 in the receptor compartment. Samples were withdrawn at predetermined time intervals for 48 h and analyzed for DPX using an Agilent 1200 HPLC system with a quaternary pump, autosampler and diode-array detector (DAD) (Agilent Technologies, Waldbronn, Germany). The separation process was carried out utilizing an Agilent Zorbax Eclipse Plus C18 column, 3.5 μ m, 4.6×250 mm (Agilent Technologies, Santa Clara, California, USA) and maintained at 40 °C. The mobile phase was acetonitrile: phosphate buffer (0.02M KH₂PO₄ adjusted to pH 3 with orthophosphoric acid) (45: 55, v/v), with a flow rate of 1 mL/min, and peaks were detected at 210 nm. The injection volume was 50 μ L. Methanol was used in the preparation of the stock solution of DPX in the concentration of 2000 μ g/mL. The range of DPX concentrations was 8–200 μ g/mL and used as working standards in the assay.

Optimization of DPX-loaded NPs formulations

Results obtained from the prepared 15 formulations, according to the design, were statistically analyzed with ANOVA and multiple response optimization using Statgraphics software. The predicted optimum formulation was prepared and evaluated and compared to the practically prepared optimum formulation for validation of the results. The optimized DPX-loaded NPs formulation was also examined for particle size, zeta potential and morphology scanning electron microscopy (SEM) and transmission electron microscopy (TEM).

Scanning electron microscopy (SEM)

The surface of the optimized DPX-loaded NPs was explored after lyophilization using SEM. The sample was mounted on metal stubs with conductive silver paint, sputtered with gold and investigated using SEM (Philips XL30; FEI, Hillsboro, OR, USA).

Transmission electron microscopy (TEM)

A sample of the optimized DPX-loaded NPs was exposed to a TEM (Model JEM-1230, JOEL, Tokyo, Japan). A drop of the diluted NPs dispersion was added onto a

microscope grid, stained with 1% phosphotungstic acid and then dried before investigation.

Pharmacokinetics evaluation on healthy human volunteers

The pharmacokinetic parameters of the optimized DPX-loaded NPs were evaluated and compared with the marketed DPX tablet, (SPIMACO, Al-Qassim, Saudi Arabia) (reference), following oral administration of a single dose (30 mg) by healthy male volunteers. An open-label, single-dose, randomized, one-period, parallel design was used in which two weeks of screening preceded 24 h study periods as previously described.²⁴ Briefly, the single-dose investigation was performed in the Egyptian Research and Development Company (ERDC), Cairo, Egypt. The study protocol was approved by the Research Ethical Committee (Protocol code: 0569/452) in agreement with the Declaration of Helsinki and the International Conference on “Harmonisation of Good Clinical Practices”. Written informed consents were obtained before volunteers were allowed to participate in the study. The prepared optimized DPX-loaded NPs and the commercial DPX tablet were administered orally (30 mg, DPX). Twelve Egyptian male volunteers (25–43 years) participated in this study and were divided into two groups (6 each). Blood samples (5 mL) were collected at 0, 0.16, 0.25, 0.5, 0.75, 1, 1.25, 1.5, 2, 2.5, 3, 4, 6, 8, 12, and 24 h and centrifuged at 3500 rpm for 10 min.

Chromatographic conditions for DPX analysis in human plasma

HPLC-MS/MS analytical method was developed at the ERDC laboratories for the determination of DPX in human plasma.²⁴ Assay linearity was verified for DPX at 2–500 ng/mL with a regression coefficient ($R^2=0.994$). The HPLC-MS/MS system consisted of HPLC Agilent series 1200 (Agilent Technologies Deutschland GmbH, Waldbronn, Germany) system with a mass spectrometer detector (Agilent 6420) Triple Quad G1311A quaternary pump, G1329A autosampler, G1322A vacuum degasser, and mass hunter software. A mixture of 75% acetonitrile: 25% buffer (ammonium formate 20 mM +0.2 % (v/v) formic acid in water) was used as a mobile phase at a flow rate of 0.35 mL/min. The reversed-phase column Intersil ODS –3 (4.6 mm x 50 cm, dp 5 μ m Sigma–Aldrich, USA) was temperature controlled at 25 °C. Sildenafil was used as an internal standard (IS).

Pharmacokinetic data analysis

Noncompartmental analysis was utilized to calculate the pharmacokinetic (PK) parameters using Kinetica® software (version 4; Thermo Electron Corp., Waltham, MA, USA). The difference between the two investigated groups in terms of drug plasma concentration was calculated by two-way ANOVA followed by Sidak's multicomparison test using GraphPad Prism 6 Software, Inc. (CA, USA). Unpaired *t*-test (two-tailed) was used to determine the significance between the pharmacokinetic parameters. The peak plasma concentration of the drug (C_{max}), the time required for drug to reach the maximum plasma concentration (t_{max}), The area under the time-concentration curve (AUC), elimination rate constant (K_{el}) and mean drug residence time (MRT) were calculated from which the relative bioavailability $[(AUC_{test}/AUC_{reference}) \times 100]$ was determined.

Results and discussions

The objective of this study was optimizing DPX-loaded NPs to improve its absorption and maximize its oral bioavailability. The predefined critical quality features of the DPX-loaded NPs were particle size, zeta potential, and DPX release pattern. Minimizing the particle size of the prepared NPs would allow for increased permeability and absorption via the mucosal membranes. Therefore, we conducted studies to optimize the factors affecting NPs

formulation to minimize the nanoparticle size, maximize zeta potential, while maximizing drug release. We then verified the efficiency of the optimized DPX-loaded NPs by conducting pharmacokinetic studies on human volunteers after oral administration of a single dose and compared the result with marketed DPX tablet.

DPX-loaded NPs formulation and characterization

Particle size and zeta potential are key parameters for evaluation of NPs stability and efficiency which influence the biopharmaceutical feature of NPs.

Particle size (Y_1)

As reported previously in several studies, the physico-chemical properties of NPs (nano-size range of the particles) rather than the drug itself, determines the release profile and absorption of the drug.^{27–29} Due to their nano-size range, NPs have large surface areas available for absorption via the epithelial surfaces.^{30,31}

Table 2 shows the variabilities in the size of the prepared NPs formulations that ranged from 148.6 to 206.8 nm for DN-8 and DN-6, with respect to changes of the investigated factors, with the results displayed in Table 3. It is worth to mention that the polydispersity index of the formulations did not exceed 0.3 which

Table 3 Statistical analysis of variance (ANOVA) of the responses (Y_1 – Y_4) results

Factors	Particle size (Y_1)		Zeta potential (Y_2)		Initial DPX release after 1 h (Y_3)		Cumulative DPX release after 24 h (Y_4)	
	Estimate	p-Value	Estimate	p-Value	Estimate	p-Value	Estimate	p-Value
X_1	43.225	0.0001*	−17.703	0.0001*	−10.06	0.0001*	−10.238	0.0032*
X_2	−2.8	0.3123	−2.002	0.0189*	0.368	0.6257	2.913	0.1923
X_3	−12.475	0.0041*	0.693	0.2903	6.588	0.0002*	2.09	0.3290
X_1^2	20.275	0.0027*	−9.704	0.0001*	−6.79	0.0013*	−4.875	0.1473
X_1X_2	−4.0	0.3079	0.842	0.3565	3.135	0.0259*	4.74	0.1435
X_1X_3	4.95	0.2192	0.041	0.9625	−5.565	0.0026*	−1.015	0.7257
X_2^2	−17.475	0.0050*	−0.810	0.3907	1.055	0.3576	3.625	0.2587
X_2X_3	7.6	0.0836	1.545	0.1214	−2.5	0.0546	−1.815	0.5361
X_3^2	−16.625	0.0062*	−1.001	0.2984	−0.645	0.5629	6.14	0.0834
R^2	98.8063		99.529		98.7042		89.9672	
Adj. R^2	96.6577		98.6813		96.3719		71.9082	
SEE	3.52484		0.828797		1.00074		2.7339	
MAE	1.75667		0.409478		0.462667		1.39367	

Note: *Significant effect of factors on individual responses.

Abbreviations: DPX, dapoxetine; ALA, alpha lipoic acid; X_1 , percentage of alpha lipoic acid in Zein; X_2 , percentage of polyvinyl alcohol; X_3 , stirring rate; Y_1 , particle size (nm); Y_2 , zeta potential (mV); Y_3 , initial DPX release after 1 h (%); Y_4 , cumulative DPX release after 24 h (%); X_1X_2 , X_1X_3 , X_2X_3 , the interaction term between the factors; X_1X_1 , X_2X_2 , and X_3X_3 are the quadratic terms between the factors; R^2 , R-squared; Adj- R^2 , Adjusted R-squared; SEE, standard error of estimate; and MAE, Mean absolute error.

indicates unimodal distribution of the size and good uniformity of the prepared formulations.

Two-way ANOVA analyses showed a significant synergistic effect of the percentage of ALA in Zein (X_1) on the particle size (Y_1) with a p -value of 0.0001. While the stirring rate (X_3) was found to have a significant antagonistic effect on Y_1 with a p -value of 0.0041. Also, it was noticed that the quadratic term of X_1 had a significant synergistic effect on Y_1 with a p -value of 0.0027. Finally, it was noticed that Y_1 was significantly affected by the quadratic term of X_2 and X_3 with p -values of 0.0050 and 0.0062, respectively. The prediction model to correlate variables with the obtained sizes is shown in Equation (4).

$$Y_1 = 116.275 - 13.5725 X_1 + 45.1 X_2 + 0.29694 X_3 + 0.4055 X_1^2 - 0.8 X_1 X_2 + 0.0025 X_1 X_3 - 34.95 X_2^2 + 0.038 X_2 X_3 - 0.0002 X_3^2 \quad (4)$$

We observed that increasing ALA percentage in the polymer content of the formulations led to a significant increase in the particle size of the NPs as observed in

the formulations DN-4, DN-6, DN-9, and DN-10. This may be explained by the formation of aggregates due to interactions between ALA and Zein matrix, consistent with a previously reported study of ALA addition to a positively charged chitosan.³² Conversely, increasing the percentage of Zein in the polymeric matrix resulted in the formation of smaller size NPs, likely due to the presence of charged amino acids in the Zein structure.^{33,34} At pH 7.2, the protein molecules aggregated into small size particles with reduced void spaces, enhancing NPs formation and hindering large aggregation via acceleration of thiol–disulfide interchange reaction.¹⁶ Another possible explanation may be related to the structural configuration of hydrophilic and hydrophobic regions on Zein surface. Zein molecules associated into an elongated prism-like shape with hydrophobic sides and hydrophilic tops and bottoms,³⁵ allowing Zein molecules to associate in a side-by-side manner to form many small NPs. This direct relationship between X_1 and the particle size (Y_1) are presented in the Pareto chart and response surface plots of Y_1 (Figures 1A and 2A, B).

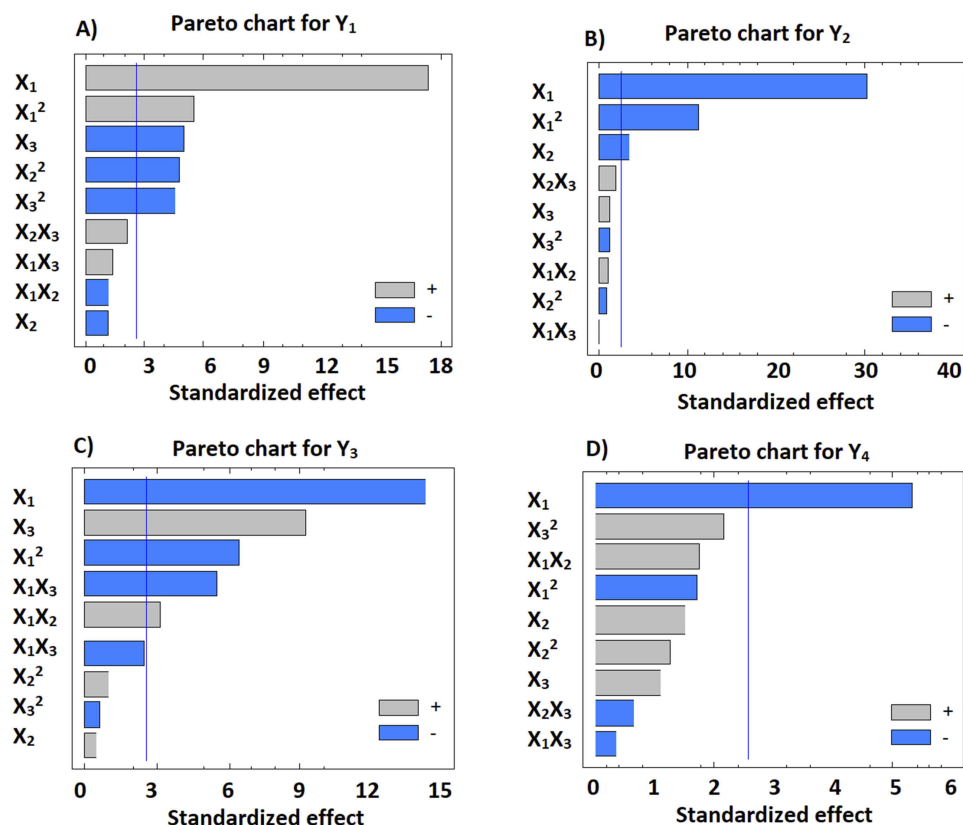


Figure 1 Standardized Pareto charts showing the significance of X_1 , X_2 and X_3 and their combined effects on Y_1 (A); Y_2 (B); Y_3 (C) and Y_4 (D).

Abbreviations: X_1 , percentage of alpha lipoic acid in zein; X_2 , polyvinyl alcohol % concentration; X_3 , stirring rate; Y_1 , particle size (nm); Y_2 , zeta potential (mV); Y_3 , initial permeation after 1 h (%); Y_4 , cumulative permeation after 24 h (%).

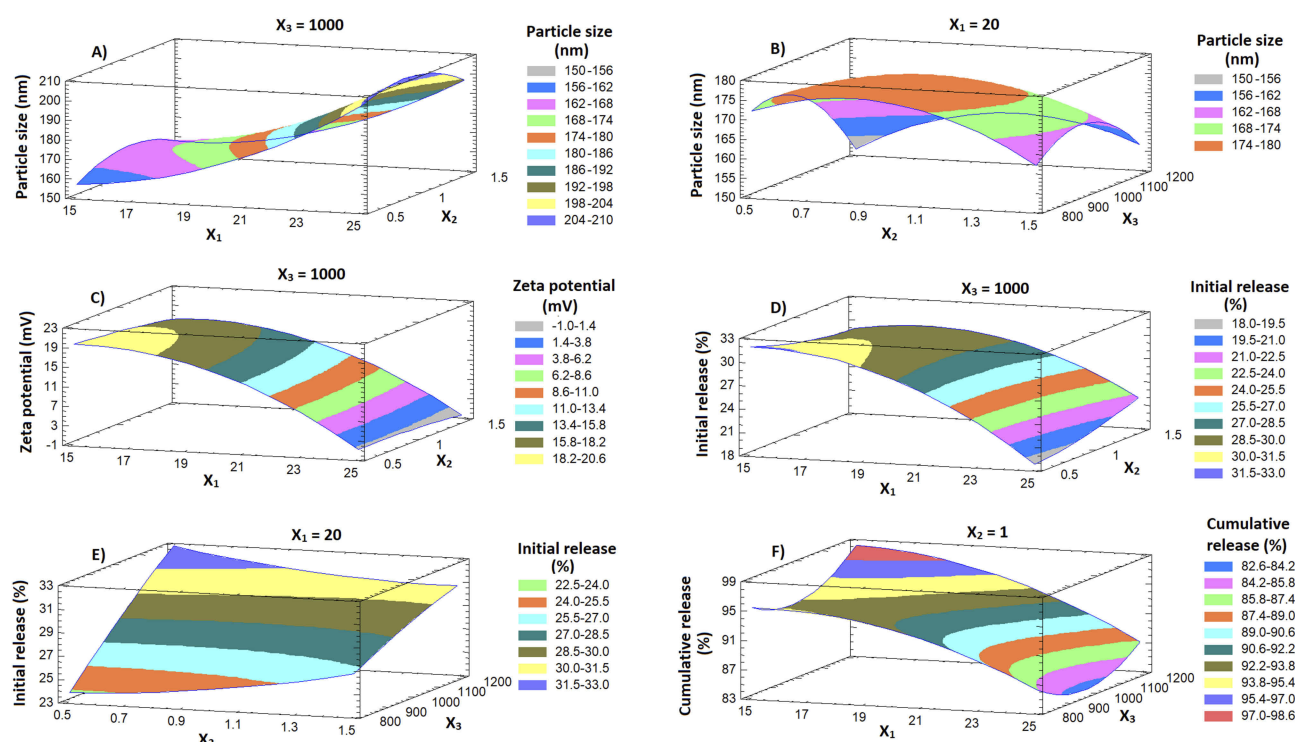


Figure 2 3D response surface plots showing the effects of X_1 , X_2 and X_3 on the investigated dependent variables Y_1 (A and B); Y_2 (C); Y_3 (D and E) and Y_4 (F).

Abbreviations: DPX, dapoxetine; X_1 , percentage of alpha lipoic acid in Zein; X_2 , polyvinyl alcohol % concentration; X_3 , stirring rate; Y_1 , particle size (nm); Y_2 , zeta potential (mV); Y_3 , initial release after 1 h (%); Y_4 , cumulative release after 24 h (%).

We observed an inverse relationship between the stirring rate (X_3) and the average particle size (Y_1). The particle size increased from 150.1 to 168.9 nm in DN-7 and DN-3, respectively when the stirring rate was decreased from 1200 to 800 rpm at an equal level of X_1 and X_2 . Also, the particle size increased from 196.9 to 205.7 nm in DN-10 and DN-4, respectively by decreasing X_3 from 1200 to 800 rpm at the same level of X_1 and X_2 . We also observed an increase of Y_1 from 148.6 to 167.3 nm for DN-8 and DN-12, respectively with a decrease in X_3 from 1200 to 800 rpm at the same level of X_1 and X_2 . This finding could be explained by increasing the shearing action and reducing the tendency of a collision between the formed NPs and decreasing the possibility for aggregation that leads to smaller particle size.^{36–38}

Zeta potential (Y_2)

Zeta potential represents a key indicator for the stability of the colloidal dispersions. The values of zeta potential of all NP formulations (Y_2) was in the range from 0.29 mV for DN-9 to 19.1 mV for DN-2 (Table 2). In general, all formulations were considerably stable except the formulations containing a high percentage of ALA (25%). It was

clear that X_1 (Percentage of ALA in Zein) and X_2 (percentage of PVA) were the main factors responsible for the difference in the zeta potential of DPX-loaded NPs as shown in Figures 1B and 2C. Multiple regression analysis indicated that both X_1 and X_2 had significant inverse effects on the zeta potential (Y_2) with a more predominant effect on X_1 with p -values of 0.0001 and 0.0189, respectively (Table 3). The prediction Equation (5) of zeta potential value is:

$$Y_2 = -30.0548 + 5.804 X_1 - 9.8487 X_2 + 0.0186 X_3 - 0.19408 X_1^2 + 0.1683 X_1 X_2 + 0.00002 X_1 X_3 - 1.6207 X_2^2 + 0.0077 X_2 X_3 - 0.00001 X_3^2 \quad (5)$$

It was evident that an increased percentage of ALA in the polymeric matrix led to markedly decreased in the zeta potential of the prepared NPs. The zeta potential value of formulations containing 15% of ALA is close to 20 mV while the formulations containing 25% of ALA is less than 1. An increase in X_1 from 15 to 25%, at the same level of X_2 and X_3 , led to a decrease in zeta potential from 17.3 to 0.29 mV for DN-1 and DN-9, respectively. The same finding was observed in DN-2 and DN-6 by decreasing the zeta potential from 19.1 to 0.41 mV, respectively. Also,

this finding could be confirmed by the decrease of Y_2 from 18.0 to 0.41 mV for DN-12 and DN-4, respectively. We attribute these observations to increased adsorption of ALA on the surface of NPs (observed with zeta potential measurement) that could neutralize the positive charges of Zein NPs. The positive charge on Zein NPs was found to decrease when the percentage of ALA increased in the polymeric matrix. The adsorption of ALA with its anionic free fatty acids on the formed cationic Zein surfaces could neutralize their positive charge.³⁹ Nishiura also explained this finding to the vicinity of the dissociated carboxyl groups of ALA that could shield the positive charge on the Zein-loaded NPs surface.⁴⁰

Despite a lesser effect of X_2 than X_1 on Y_2 , it was found that increasing X_2 from 0.5 to 1.5 %, at the same level of X_1 and X_2 , led to the decrease in zeta potential from 19.1 to 17.3 mV for DN-2 and DN-1, respectively. The same finding was observed in DN-7 and DN-11 by decreasing the zeta potential from 15.1 to 13.6 mV, respectively. Also, this finding could be confirmed by the decrease of Y_2 from 15.2 to 10.6 mV for DN-3 and DN-5, respectively. This observation could be attributed to incomplete hydrolysis of the starting material resulting in a number of acetic groups that are capable of ionizing and reducing the positive charge that affects zeta potential of the prepared nanoparticles.⁴¹

DPX EE %, DPX loading and yield percentage
The data for DPX EE% showed no significant change in EE% among the prepared formulations with an average value of $34.35 \pm 4.66\%$. DPX is mainly charged at physiological pH with its pKa of 8.6. The protonation of DPX is attributed to the elevated aqueous solubility (68 mg/mL) that contributes to the reduction of EE %. In addition, results revealed that the prepared formulations showed average DPX loading % and yield % of 18.9 % and 54.52%, respectively.

In vitro release behavior of DPX from NP formulations (Y_3 and Y_4)

This study was conducted to predict the in vivo behavior of the nanoparticle. The in vitro release of DPX from the prepared NPs formulations (DN-1 to DN-15) showed a biphasic form with an initial rapid release followed by a controlled release phase (Figure 3). The initial release after 1 hr range from 18.09% with DN-6 to 35.67% with DN-8 (Table 2). Whereas, the cumulative % of DPX released after 24 hrs showed DN-2 to have the most release (97.65%)

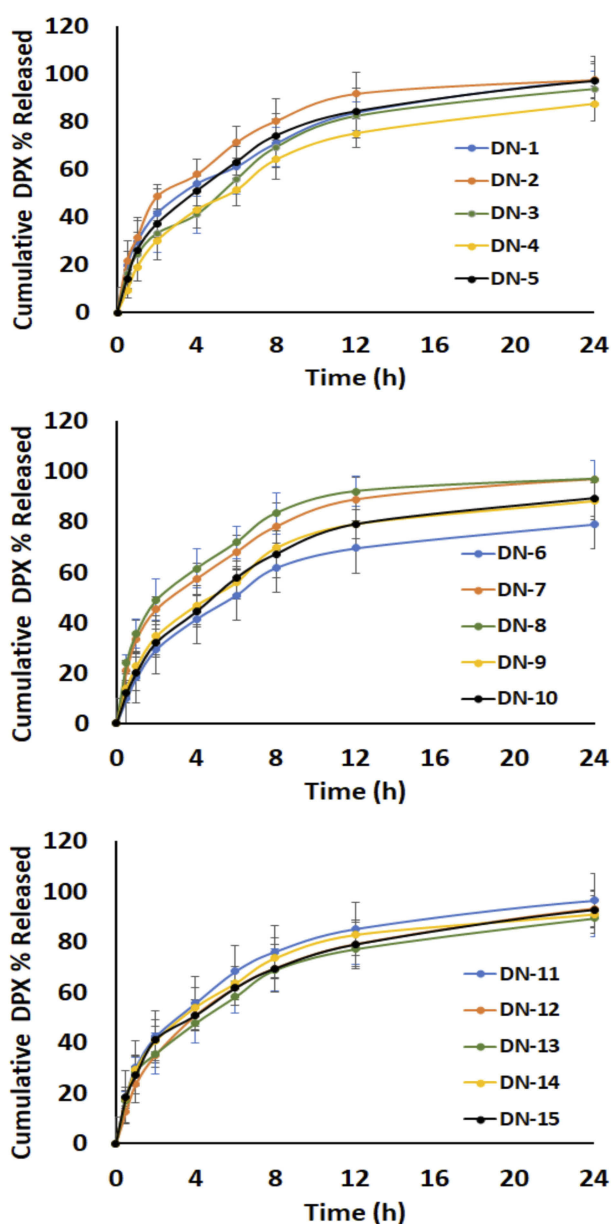


Figure 3 In vitro release profiles of DPX-loaded NP formulations: (A) F1-F5; (B) F6-F10; and (C) F11-F15.

Abbreviations: DPX, dapoxetine; ALA, alpha lipoic acid.

with DN-6 showing the least release (79.05%) (Table 2). Regression analysis with ANOVA detected that both Y_3 and Y_4 are governed by the percentage of ALA in Zein (X_1), which had a significant antagonistic effect on Y_3 , and Y_4 with p -values of 0.0001, and 0.0032, respectively (Table 3 and Figure 1C, D and 2D-F). Also, the stirring rate (X_3) was found to have a significant antagonistic effect on the initial DPX release percentage after 1 h (Y_3) with a p -value of 0.0002. Also, it was noticed that the quadratic term of X_1 had a significant antagonistic effect on Y_3 with a p -value of 0.0013. Finally, it was

also noticed that the initial DPX release percentage after 1 h (Y_3) was significantly affected by the interaction terms X_1X_2 and X_1X_3 with p -values of 0.0259 and 0.0026, respectively. The equations (6 and 7) predicted the responses Y_3 and Y_4 :

$$Y_3 = -84.2788 + 6.5815 X_1 - 3.8925 X_2 + 0.1007 X_3 - 0.1358 X_1^2 + 0.627 X_1X_2 - 0.00278 X_1X_3 + 2.11 X_2^2 - 0.0125 X_2X_3 - 0.000008 X_3^2 \quad (6)$$

$$Y_4 = 148.252 + 2.4358 X_1 - 21.4725 X_2 - 0.12905 X_3 - 0.0975 X_1^2 + 0.948 X_1X_2 - 0.0005 X_1X_3 + 7.25 X_2^2 - 0.009 X_2X_3 + 0.00008 X_3^2 \quad (7)$$

Data obtained from the release profiles indicate (Table 2 and Figure 3) that, the percentage of ALA increased in the polymeric matrix (X_1) significantly impacted both the initial and cumulative DPX release. An increase in X_1 from 15 to 25%, at the same level of X_2 and X_3 , led to a decrease in Y_3 from 29.68 to 22.7% for DN-1 and DN-9, respectively. Similar observation found in DN-2 and DN-6 where Y_3 decreased from 31.34 to 18.09%, and 23.54 to 19.1% for DN-12 and DN-4, respectively. The same formulations revealed a similar trend regarding the cumulative DPX release (Y_4). This behavior can be explained by the release of surface bound DPX from Zein-loaded NPs during the initial rapid release phase. DPX diffusion from NPs core is responsible for the controlled release phase. In addition, the hydrophobic nature of Zein and ALA NPs can sustain the encapsulated DPX release.

Moreover, it was found that there is a direct relationship between the stirring rate (X_3) and the initial DPX release (Y_3). The initial DPX release decreased from 20.1 to 19.1% in DN-10 and DN-4, respectively when the stirring rate was decreased from 1200 to 800 rpm at an equal level of X_1 and X_2 . Also, Y_3 decreased from 35.67 to 23.54% in DN-8 and DN-12, respectively by decreasing X_3 from 1200 to 800 rpm at the same level of X_1 and X_2 . We observed a similar decrease of Y_3 from 33.45 to 24.34% for DN-7 and DN-3, respectively due to the decrease in X_3 from 1200 to 800 rpm at the same level of X_1 and X_2 . This finding may be due to the size reduction of the NPs by increasing the stirring rate, which permits greater surface area in contact with the dissolution media that promotes penetration of NPs, consequently improving the drug dissolution and increasing the initial drug release.

Validation of optimized DPX-loaded NPs

Box-Behnken experimental design deduced the optimum DPX-loaded NPs formulation that was prepared and evaluated. The obtained results indicated that combination of independent factors for the optimized DPX-loaded NPs formulation showed the desired particle size of 159.24 nm with polydispersity index value of 0.140 (Figure 4A) and zeta potential of 19.14 mV (Figure 4B). In addition, the optimized formulation releases 25.32% of its DPX content in the first hour and 95.89% after 24 hrs. The results of the observed and the predicted values are shown in Table 4 that demonstrated no considerable residuals, and the predicted error percentage of the responses were less than 6%.⁴²

Morphological examination of the optimized DPX-loaded NPs

We undertook a morphological examination of the optimized DPX-loaded NPs to confirm the spherical morphology of the optimized NPs. SEM photomicrographs at different magnification (Figure 4C and D) revealed the spherical shape with the smooth surface of the optimized DPX-loaded NPs. The mechanical stress of ice crystals during the lyophilization process could be behind the aggregation of the NPs displayed in the images.⁴³ TEM images (Figure 4E and F) confirmed the spherical structure with a dark core of the NPs. No aggregation was observed, and the size of the NPs was in good agreement with that obtained employing the dynamic light scattering technique. The results suggest uniformity of size distribution of the prepared NPs.

Single-dose clinical pharmacokinetics investigation of the optimized DPX NPs formulation on healthy human volunteers

To study the bioavailability of DPX-loaded NPs, human volunteers were given single oral administration of 30 mg of DPX-loaded NPs. The results presented in Figure 5 and Table 5 show a significant difference ($p < 0.05$) of t_{max} , AUC₍₀₋₂₄₎ and AUC_(0-∞) for optimized DPX-loaded NPs when compared with the marketed DPX tablet. DPX-loaded NPs significantly prolonged t_{max} by 4-fold (2.5 ± 0.314 h vs. 0.583 ± 0.144 h, respectively), AUC by 2-fold, with a slight reduction of $t_{1/2}$ (5.283 ± 1.077 h vs. 8.452 ± 2.813 h). On the other hand, the data showed no significant difference ($p < 0.05$) in the C_{max} values of the NPs formulation compared with the marketed DPX tablet.

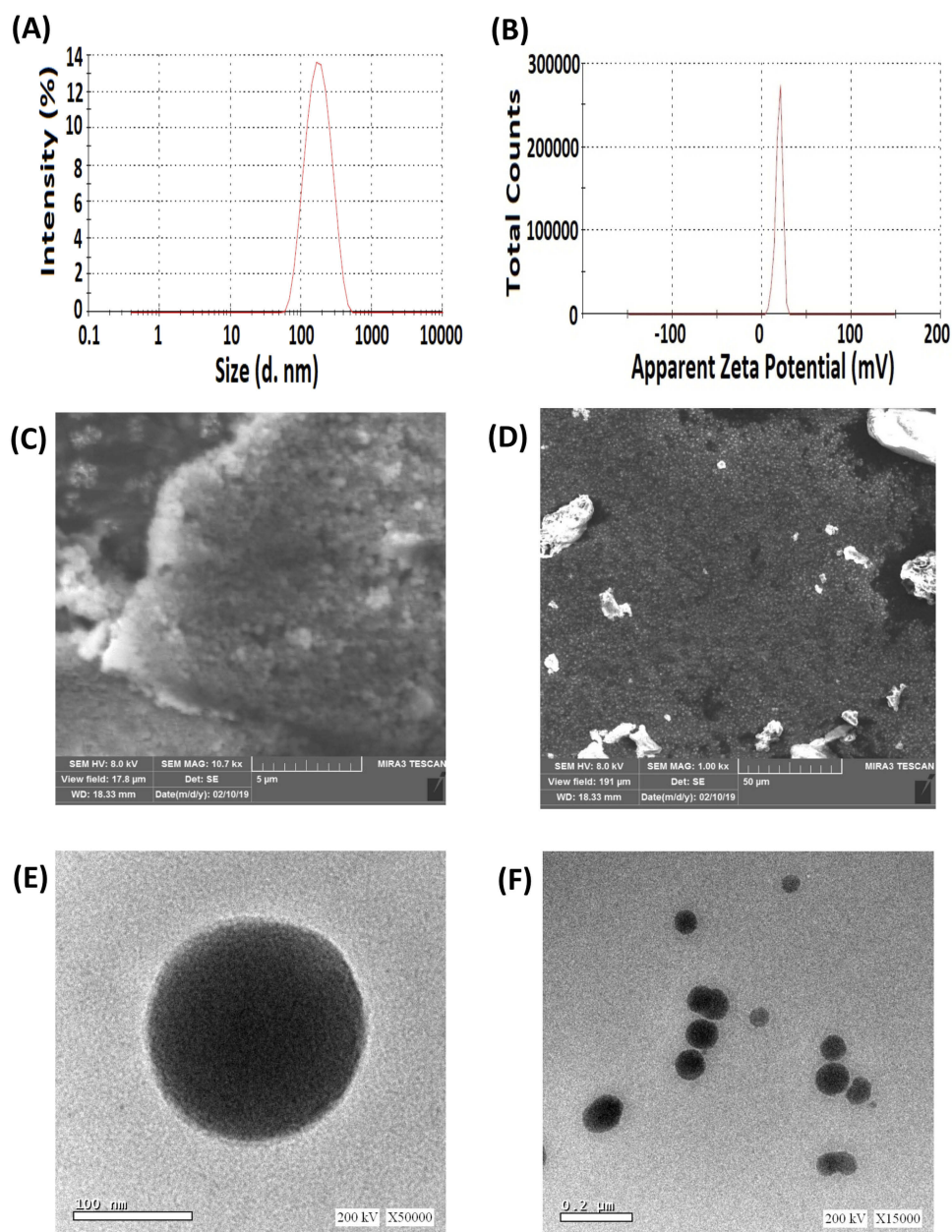


Figure 4 Particle size distribution (A), zeta potential (B) measured by particle size analyzer, SEM photomicrographs (C and D) and TEM images (E and F) of optimized DPX-loaded NPs.

Abbreviations: TEM, Transmission electron microscope; DPX, dapoxetine; ALA, alpha lipoic acid.

Most importantly, the optimized DPX-loaded NPs formulation showed 194% improved relative bioavailability when compared to the marketed tablet. The reduction of DPX half-life ($t_{1/2}$) in the DPX-loaded NPs is most likely due to the improved permeability, absorption and distribution of DPX, from the NPs formulation to the second compartment tissue. This could be related to reduced particle size (to the nanoscale range) formulation that leads to enhanced DPX oral absorption with rapid elimination. The

enhancement in the bioavailability is likely due to the prolongation in the MRT of the NPs to 7.637 ± 1.373 h compared to 6.031 ± 1.826 h for the marketed tablet. Prolongation in MRT was accompanied by prolongation in the t_{max} from 0.583 ± 0.144 h of the tablet to 2.5 ± 0.314 h of the NPs. Prolongation of the t_{max} and MRT of the DPX in the NPs could be due to the controlled release behavior of the encapsulated DPX in the NPs. Also, the presence of Zein and ALA in the formulation may be sterically

Table 4 Optimal calculated variables and observed, predicted and residual values for responses (Y_1 – Y_4)

Factor	Optimum	Response	Predicted	Observed	Residual
X_1 (%)	15	Y_1 (nm)	161.88	159.24	2.64
X_2 (%)	0.5	Y_2 (mV)	19.49	19.14	0.35
X_3 (rpm)	800	Y_3 (%)	24.22	25.32	1.10
		Y_4 (%)	97.19	95.89	1.29

Abbreviations: X_1 , percentage of alpha lipoic acid in zein; X_2 , polyvinyl alcohol % concentration; X_3 , stirring rate; Y_1 , particle size (nm); Y_2 , zeta potential (mV); Y_3 , initial permeation after 1h (%); Y_4 , cumulative permeation after 24 h (%).

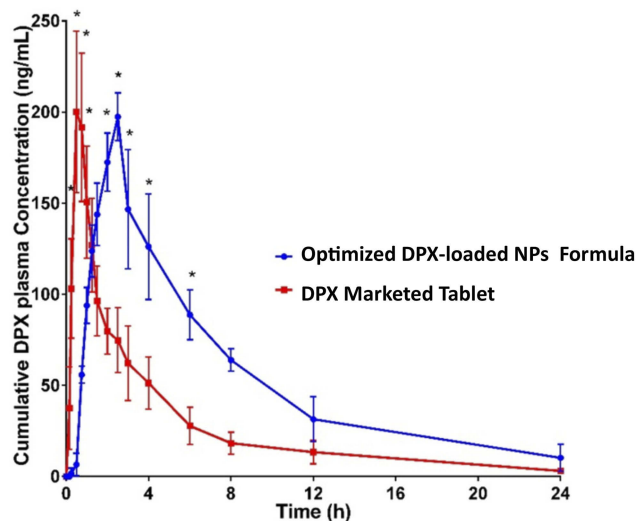


Figure 5 Plasma concentration–time profiles for the optimized DPX-zein-ALA nanospheres and marketed DPX tablets (30 mg) administered orally as a single dose. Data represent the mean value $\pm$ standard deviation ($n=6$). *Significant at $P<0.05$ (two-way ANOVA, Sidak's multi-comparison test).

hindering the uptake of NPs by the tissue through the reticuloendothelial system that decreases its rate of clearance from the body.^{29,44} In addition, the N-terminal region for Zein protein interacts with cell membranes and serve as a peptide carrier for drugs and thus enhances paracellular permeability.⁴⁵ These features underpin the enhanced bioavailability of the developed DPX-loaded NPs formulation.

Conclusion

The introduction of DPX for the treatment of PE eliminates the reliance on other non-approved treatments, however, it is fraught with low bioavailability leading to a high therapeutic dose and adverse effect. We have successfully developed an optimized DPX-Zein-ALA formulation that has resulted in the improvement of oral bioavailability of DPX in human volunteers when compared to the marketed

Table 5 Single dose clinical pharmacokinetic parameters after oral administration of 30 mg of optimized DPX-loaded NPs and DPX marketed tablets

Pharmacokinetic Parameter	Optimized DPX-loaded NPs	DPX Marketed Tablets
C_{max} (ng/mL)	197.384 $\pm$ 13.153	200.121 $\pm$ 44.351
t_{max} (h)	2.5 $\pm$ 0.314*	0.583 $\pm$ 0.144
AUC _(0–24) (ng.h/mL)	1290.375 $\pm$ 262.169*	662.758 $\pm$ 116.011
AUC _(24–∞) (ng.h/mL)	85.77 $\pm$ 79.191	46.419 $\pm$ 31.879
AUC _(0–∞) (ng.h/mL)	1376.145 $\pm$ 339.592*	709.178 $\pm$ 146.307
AUMC _(0–24) ng.hr ² /mL	8749.718 $\pm$ 2815.176	3324.778 $\pm$ 1226.219
AUMC _(24–end) ng.hr ² /mL	2058.471 $\pm$ 1900.592	1114.068 $\pm$ 765.084
AUMC _(0–∞) ng.hr ² /mL	10,808.189 $\pm$ 4708.753	4438.846 $\pm$ 1991.228
K_{el} (h ^{−1})	0.135 $\pm$ 0.025	0.089 $\pm$ 0.032
$t_{1/2}$ (h)	5.283 $\pm$ 1.077	8.452 $\pm$ 2.813
MRT (h)	7.637 $\pm$ 1.373	6.031 $\pm$ 1.826
Relative bioavailability (%)	194.04	–

Notes: Data represent the mean value $\pm$ standard deviation (SD) ($n=6$), *Significant at $P<0.05$ unpaired t test (two-tailed).

DPX tablet. These results revealed the ability of the NPs formulation to improve the delivery and effect of DPX that could be applied successfully in the management of male PE.

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Disclosure

The authors report no conflicts of interest to declare in this work.

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