

Enhancement of Oral Bioavailability of Ibandronate Through Gastroretentive Raft Forming Drug Delivery System: In Vitro and In Vivo Evaluation

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Background: Bisphosphonates have very low bioavailability and cause irritation of the esophagus and stomach. This study was planned to improve the oral bioavailability of ibandronate through the formation of a raft in the stomach. Bisphosphonate-induced irritation of the esophagus and stomach is prevented by the formation of a raft.

Materials and Methods: The nanostructured raft was developed through the use of nanosized citrus pectin (NCP). The particle size of NCP was measured by zeta sizer and SEM. The percentage of NCP and the neutralization profile of raft was studied. The ibandronate, polymers, and the developed formulation were characterized by FTIR, XRD, TGA, and DSC. The release of ibandronate was studied in 0.1 N HCl, 0.5 N HCl, 1 N HCl, and simulated gastric fluid (SGF) and a cell viability study was performed using Caco-2 cells. The PPR5 formulation and Bonish 150 mg tablets were selected as test and reference formulations, respectively, for pharmacokinetic study. Twelve healthy albino rats were taken and divided into two groups using a Latin square crossover design, and the blood samples were collected for 24 hours.

Results: The SEM image showed that the particle size of NCP was 159 nm. The raft of PPR5 showed 94% NCP and 45 minutes duration of neutralization. The FTIR and XRD showed chemical stability and a uniform distribution of ibandronate in the raft. The TGA and DSC indicated the thermal stability of formulation. The release of 99.87% ibandronate at 20 minutes was observed in the SGF. The values of C_{max} for the reference and test formulations were 493 ± 0.237 ng/mL and 653 ± 0.097 ng/mL, respectively. The $AUC_{(0-1)}$ of the reference and test formulations was 3708.25 ± 3.418 ng/mL.h and 6899.25 ± 3.467 ng/mL.h, respectively.

Conclusion: The NCP has been successfully prepared from citrus pectin and has shown effective porous raft formation. The bioavailability of the ibandronate from newly developed PPR5 was higher than the already marketed formulation.

Keywords: nanosized citrus pectin, raft, in vitro release, cell viability, pharmacokinetics

Introduction

Bisphosphonates are used in the management of a variety of diseases related to Paget's disease, malignancy induced hypercalcemia, post-menopausal osteoporosis, and bone resorption. For drugs belonging to BCS class III (high solubility and low permeability), achieving the desired bioavailability is quite difficult due to the low permeability of the drug through gastrointestinal mucosa. Ibandronate, a famous

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bisphosphonate belonging to the BCS class III used in therapy for the treatment of osteoporosis, has very low bioavailability when administered orally in the form of conventional tablets.¹ Oral bisphosphonate therapy, including ibandronate linked to upper gastrointestinal injury, has unfortunately raised concerns. Reports of adverse experiences have shown that oral bisphosphonate can potentially cause severe intolerance to the upper gastrointestinal tract due to irritation of the local mucosa, particularly to produce unwanted esophageal experiences, such as esophagitis, ulcers, and erosions in a small percentage of patients.

In recent decades, polymers have been extensively used as raw constituents for proposed drug delivery systems due to their excellent properties, such as non-toxicity, biocompatibility, biodegradability, and environmental sensitivity, etc. Pectin is one of the most abundant natural biopolymers, which is isolated and purified from citrus peel.² This polymer is insoluble in water and has good film-forming properties. Pectin has gained many attractions in the pharmaceutical and medical applications. Due to its exclusive antacid and antiulcer features as well as its gelling and film forming properties, pectin has a great prospective for use as a suitable carrier in drug delivery systems. The pectin is processed in the laboratory by an appropriate method to convert it to NCP, which is then used in the study for the formation of the effective porous nature of the raft.

The raft forming delivery system is formulated containing NCP as the raft former, sodium bicarbonate as effervescent mixture, calcium carbonate to enhance the raft strength, and polyethylene glycol (PEG 400) as the permeability enhancer, while it is necessary to mention that the absorption window of ibandronate is located inside the stomach and the raft forming system ensures the availability of drug at gastric mucosa for a relatively prolonged time period. The addition of the permeability enhancer and prolonging the residence time of the delivery system at its absorption window increased the bioavailability of ibandronate. The purpose of this research work was to enhance the oral bioavailability of ibandronate through a raft forming delivery system. The NCP and PEG 400 were used as raft former and penetration enhancer, respectively. The developed raft was evaluated by FTIR, XRD, TGA, DSC, and cell viability study. The pharmacokinetic of ibandronate was studied on albino rats.

Materials and Methods

Chemicals and Reagents

Ibandronate was gifted by Global Pharmaceuticals Pvt. Ltd. Islamabad, Pakistan. citrus pectin, PEG 400, cross-linked carboxymethyl cellulose (CCMC), potassium dihydrogen phosphate, acetonitrile, sodium chloride and ethanol were purchased from Sigma-Aldrich GmbH, Germany. Calcium carbonate, citric acid, sodium bicarbonate and pectin esterase (PE) were obtained from Merck Darmstadt, Germany. Bonish (150 mg tablets) was purchased from a local market.

Preparation of Nanosized Citrus Pectin (NCP)

The NCP was prepared by the method already reported by Jiang et al,³ in 2001. Only a 1% solution of citrus pectin was prepared using 0.4 M NaCl solution and adjusted the pH to 6.5 by 0.1 N NaOH solution. Then 6 mL of citrus pectin solution was mixed with an equal volume of 0.2 M phosphate buffer pH 6.5 and centrifuged at 15,000 rpm for 10 minutes. The supernatant was obtained and filtered through a 0.45 μ m filter. The obtained citrus pectin solution was mixed with 100 units/mL of pectin esterase (PE). The PE-citrus pectin solution was incubated at 30°C for 16 hours and the solution was placed in a water bath to stop the reaction for 3 minutes. The solution was dried in an oven at 45°C for 4 hours. The particle size of the dried NCP was determined by a particle size analyzer and SEM.

Preparation of Raft Forming Tablets

The tablets for prompt delivery of ibandronate were prepared by wet granulation method.⁴ The ibandronate, NCP, PEG 400, and other excipients were mixed thoroughly, as shown in Table 1. The powdered mixture of ibandronate and other excipients was mixed using a Sigma mixer. The powdered mixture was granulated using 2 % (w/w) HPMC E5 in a 90% solution of ethanol. The granules were dried at 40°C for 2 hours, passed through 18-mesh screen and compressed using a minipress MII (Pharma Test, Hainburg, Germany).

Disintegration Time of Tablets

One tablet was placed in a 250 mL beaker containing 120 mL of water at room temperature to measure the disintegration time. The tablet was disintegrated if the evolution of gas around the tablet or its fragments stopped, being either dissolved or dispersed in water so that no agglomerate remains. The process was repeated on four further tablets.

Table I Composition of NCP Raft Forming Formulations

| Code | Ibandronate (%) | NCP (%) | CCMC (%) | PEG 400 (%) | NaHCO ₃ (%) | Citric acid (%) | CaCO ₃ (%) |
|------|-----------------|---------|----------|-------------|------------------------|-----------------|-----------------------|
| PPR1 | 30.00 | 12.50 | 2.00 | 2.00 | 28.00 | 14.00 | 11.50 |
| PPR2 | 30.00 | 27.50 | 2.00 | 2.00 | 19.00 | 9.50 | 10.00 |
| PPR3 | 30.00 | 10.00 | 2.00 | 2.00 | 30.00 | 15.00 | 11.00 |
| PPR4 | 30.00 | 12.50 | 2.00 | 2.00 | 30.00 | 15.00 | 8.50 |
| PPR5 | 30.00 | 10.00 | 2.00 | 2.00 | 32.00 | 16.00 | 8.00 |
| PPR6 | 30.00 | 2.50 | 2.00 | 2.00 | 36.00 | 18.00 | 9.50 |
| PPR7 | 30.00 | 2.50 | 2.00 | 2.00 | 39.00 | 19.50 | 5.00 |
| PPR8 | 30.00 | 7.50 | 2.00 | 2.00 | 37.00 | 18.50 | 3.00 |
| PPR9 | 30.00 | 5.00 | 2.00 | 2.00 | 39.00 | 19.50 | 2.50 |

Hardness of Tablets

This was measured because, during preparation, transportation, and storage of tablets, the dosage forms faces the mechanical shock. Much equipment is available for the measurement of hardness of tablets. The hardness of tablets was measured in Kg/cm² by using Erweka hardness tester (Gmbh Germany).

Properties of NCP Raft

Determination of NCP in the Raft

The percentage of NCP in the raft was determined by the previously established HPLC method. Then 150 mL of SGF was placed in a beaker and the temperature was maintained at 37°C. The tablet was put in a beaker and we waited until the raft was completely formed. The raft was transferred to the Büchner funnel using a spatula, and washed with purified water (n=3). The raft was transferred to a centrifuge tube with the help of a spatula and a sufficient amount of purified water was added. Ethanol was added at 4°C to a centrifuge tube and centrifuged for 5 minutes at 5000 rpm. The raft was placed in an oven at 40°C after removing it from the centrifuge tube. The fine powder of the raft was made by grinding, placed in a centrifuge tube, 5 mL of phosphate buffer at pH 7.4 was added and centrifuged for 5 minutes at 5000 rpm. The upper clear layer was removed, filtered through 0.22 µm filter, and injected into an HPLC system for analysis. The following equation was used to calculate the contents of the NCP.

$$\text{Contents of NCP} = (\text{Xmg/mL}) \times \frac{V(\text{mL}) \times \text{total raft weight}(\text{mg})}{\text{weight of sample}(\text{mg})} \quad (1)$$

Acid Neutralization Capacity (ANC) Within the Raft

The acid neutralization capacity of the raft was measured to understand the ability of the raft to retain the antacid and to provide an antacid reservoir with a benefit. The two flasks were taken with a capacity of 500 mL. De-ionized water and 1 M HCl was added to flask one and it was incubated at 37°C for 20 minutes using a water bath, which was reagent A; 0.5 M NaOH was added to the second flask and heated for 20 minutes at 37°C, which was reagent B. The tablet was placed in a beaker and left until the raft was completely formed. The raft was transferred to a centrifuge tube after washing with purified water (n=3). A sufficient amount of purified water was added, with ethanol at 4°C, and it was centrifuged for 5 minutes at 5000 rpm. The raft was placed in an oven at 40°C after removing it from the centrifuge tube. Powder of the raft was placed in a conical flask with 150 mL of purified water.⁵ The flask was placed over a shaker at 250 rpm for 60 seconds with 30 mL of reagent A. With a burette, reagent B was added and the titration analysis started. The pH meter was used to check the changes in pH. The following equation was used to calculate the ANC.

$$\text{ANC} = V - T \times 0.5 \times \frac{\text{total mass of raft}(\text{mg})}{\text{weight of sample}(\text{mg})} \quad (2)$$

where V is the volume (mL) of HCl, and T is the volume (mL) of titre.

Profile of Neutralization

The neutralization profile of the raft is used to measure the capacity for the raft to act as a reservoir of antacids and provide sustained protection against gastric acid of stomach; 150 mL of SGF were taken in a beaker and the temperature was maintained at 37°C. The tablet was placed in a beaker

and left until the raft was completely formed. The developed NCP raft was transferred to a Büchner funnel and any media were discarded. It was left for 5 minutes and then 3 mL of 0.04 M HCl was added on the raft. The obtained solution was discarded and 0.1 M HCl solution was placed on the raft for filtration ($n=3$), the solution was collected, and pH was recorded by pH meter (digital).⁵

Swelling of NCP Raft

Then 150 mL of SGF were taken in a beaker and the temperature was maintained at 37°C. The tablet was placed in a beaker and left until the raft was completely formed. The raft was transferred to the Büchner funnel using a spatula, and left until the supernatant was completely removed. The raft was transferred to the electronic weight balance with the help of a spatula and weighed (W_0). A plastic container was taken having a solution of 0.1 N HCl, the raft was added in it with a mesh and placed on a shaker (orbit). The raft was removed from the plastic container after 10, 20, 30, 40, 50, and 60 minutes, the extra water was removed, and it was weighed (W_1). The raft was placed in an oven at 80°C and dried until constant weight and the final weight was recorded (W_2).⁶ The swelling was calculated using the following equation.

$$\% \text{ Swelling of pectin raft} = \frac{W_1 - W_2}{W_0} \times 100 \quad (3)$$

Floating Lag Time (FLT) Total Floating Time (TFT)

FLT and TFT of the NCP raft were determined by using USP dissolution apparatus II (pharma test Hainburg, Germany) using 900 mL SGF pH 1.2 maintained at $37 \pm 0.5^\circ\text{C}$ and at 50 rpm. The time required for the raft to rise to the surface and float was determined as FLT. TFT is the total time for which the raft floats in dissolution medium including FLT.

Characterization

The XRD diffractograms, FTIR spectra, TGA, and DSC thermograms of ibandronate, NCP, CCMC, and the raft of PPR5 were obtained using a diffractometer, FTIR-spectrophotometer (Bruker Alpha, Germany), TGA analyzer, and differential scanning-calorimeter DSC-60 (Shimadzu, Germany) respectively. The FTIR spectra was recorded in the wavelength range of 4000 to 1000 cm^{-1} . The samples for TGA and DSC was placed in an aluminium pan and analyzed under a 100 mL/min gas stream of nitrogen and heating rate was 50 to 400°C .

Ibandronate Release from Nanostructured Raft

The release of ibandronate from raft forming tablets was determined by a method already reported by Jang et al⁷ in 2014. The tablet was placed in 50 mL of four different acidic mediums, ie, SGF, 0.1, 0.5, and 1.0 N HCl solution in 250 mL capacity beaker. The temperature was maintained at 37°C and 2 mL of sample was collected at 5, 10, 15, 20, 25, 30, 45, and 60 minutes. The same volume of fresh dissolution medium was added into the beaker containing tablets for dissolution study to maintain the sink condition. The samples obtained at regular time intervals were filtered through a $0.22 \mu\text{m}$ filter and injected into the HPLC system for analysis. The concentration of ibandronate in tablets was analyzed using a HPLC system with UV detection (PerkinElmer, Shelton, CT,) with RHS C_{18} column ($1.5 \text{ cm} \times 4.6 \text{ mm}$, $5 \mu\text{m}$) (Agilent technologies, Santa Clara, CA). The analysis was performed using a mobile phase consisting of acetonitrile and 0.01 M sodium dihydrogen phosphate in a ratio of 20:80% v/v at a flow rate of 1 mL/min. The pH of the mobile phase was adjusted to 3.3 using 10% ortho-phosphoric acid. The detection wavelength was set at 195 nm. The retention time of ibandronate was 4.598 ± 0.1 minutes. The linearity parameter was studied in the range of 200 to 800 ng/mL with a correlation coefficient (R^2) of 0.999. Percentage recovery was calculated as 100.7%. The percentage relative standard deviation of precision value was found to be less than 2%.

Cell Viability Study (Commercially Purchased Cells)

Cell viability was done by MTT assay using the Caco-2 model to evaluate the cellular toxicity of the NCP raft. The Caco-2 cell line was purchased commercially (American Type Culture Collection, ATCC, Manassas, VA). Briefly, the Caco-2 cells were seeded in 96-well plates in Eagle's Minimum Essential Medium (Catalog No. 30–2003) with 20% fetal bovine serum (FBS). After confluence, the cells were incubated with DMEM without FBS and having 0.5% dispersions of various samples for 6 and 24 hours. After incubation, the samples were removed thoroughly and washed three times with phosphate buffered saline (1X PBS). Then, 500 μL MTT solutions in FBS-free medium (0.5 mg/mL) were added to each well and the cells were incubated for 1 hour. After removal of the supernatants, the converted dye was solubilized with 500 μL of DMSO. The absorbance of the resulting solution was recorded

immediately at $\lambda=570$ nm. The percentage of cell viability was calculated using the following equation;

$$\text{Cell viability(\%)} = \frac{A_s}{A_d} \times 100 \quad (4)$$

A_s and A_d are the absorbance measured after treatment with sample dispersions and DMEM, respectively.

Stability Studies

The stability studies were carried out according to ICH guidelines for 6 months by storing the PPR5 formulation in a stability chamber at 40°C and 75±5% relative humidity.⁸ The samples of optimized formulation were evaluated for disintegration time, hardness of tablets, FLT, TFT, % drug content, and % release profile of ibandronate after 1, 3, and 6 months.

Pharmacokinetic Analysis

With the approval of the Ethical Committee for Utilization of Laboratory Animals, Government College University Faisalabad, Pakistan, healthy albino rats weighing 400–600 g were divided into two groups (control and test group) having six rats in each. The ICH guidelines were followed for the handling and welfare of animals during the pharmacokinetic study. The animals were raised under normal conditions with free access to standard diet and water, housed in a well maintained room at 25±5°C with a 12:12 hour light–dark cycle. Animals were fasted for 24 hours before starting the experiment but allowed free access to water. A single dose of Bonish 150 mg tablets (equivalent to 1 mg/kg, control group) and the PPR5 (equivalent to 1 mg drug/kg, test group) was administered orally through a feeding tube followed by rinsing with water. All animals were tagged properly and retained in the wooden boxes during the sampling procedure. At pre-dose ($t=0$), and at 0.5, 1, 2, 3, 4, 6, 8, 12, and 24 hours after administration, a 0.25 mL blood sample was collected from the tail vein of the rat into heparinised micro-centrifuge tubes. Plasma samples were prepared by centrifugation of the blood (5500×g for 10 minutes), aspiration into cryo-vials and stored at –20 °C. A liquid–liquid extraction was performed to obtain ibandronate from plasma samples. To 100 μ L aliquot of plasma, 100 μ L of acetonitrile was added, vortexed for 20 minutes and centrifuged (Hermile Z 220-A) at 4000 rpm for 25 minutes. After centrifugation, an organic layer was withdrawn by using a micropipette and solvent was dried under a gentle stream of nitrogen at 45°C. The residue was reconstituted with 100 μ L of the mobile

phase and analysed by high performance liquid chromatography–UV spectrometry.

Data Analysis

The concentrations of ibandronate in plasma samples were calculated based on the calibration curve over the range of 200–800 ng/mL using Microsoft® Office Excel 2010. Pharmacokinetic parameters were estimated by using scientific application package Kinetica R version 4.1.1 (Thermo Electron Cooperation, USA). Values for maximum plasma concentration (C_{max} , ng/mL) and time to reach the peak plasma concentration (t_{max} , h) were determined directly from the average obtained data using Equations (5) and (6). Area under the plasma concentration time curve to time t (AUC_{0-t} , ng/mL.h) and area under the plasma concentration time curve to infinity ($AUC_{0-\infty}$, ng/mL.h) were calculated by mixed log-linear method using Equation (7). AUMC was calculated using Equation (8). Elimination half-life (k_{el} , h^{-1}), Clearance (Cl , h^{-1}), and mean residence time (MRT, h) were calculated using Equation (9).

$$C_{max} = \frac{FX_0}{V_d} \times e^{-kt_{max}} \quad (5)$$

$$t_{max} = 2.303 \log (K_a/K_e)(K_a - K_e) \quad (6)$$

$$AUC_{0-t} = \sum_1^n \frac{C_i + C_{i+1}}{2} \cdot \Delta t \quad (7)$$

$$AUMC_{0-t} = \sum_1^n \frac{t_i(C_i + C_{i+1})}{2} \cdot \Delta t \quad (8)$$

$$MRT = \frac{AUMC}{AUC} \quad (9)$$

where F , V_d , and k are administered dose fraction, volume of distribution, and rate constant, respectively. K_a and K_e are rate constants for absorption and elimination, respectively. Δt (t_2-t_1), C_i , and C_{i+1} are time interval, initial amount, and final amount of the drug, respectively.

Statistical Analysis

The pharmacokinetic data was statistically analyzed using SPSS version 17 (SPSS Inc., Chicago, IL). The software program SPSS (SPSS 17) Student's t -test was performed to compare the parameters between the control group and test group, with confidence interval at 95%.

Results and Discussion

Disintegration and Hardness of Tablets

The disintegration time of nine formulations ranged from 23 ± 0.02 – 56 ± 0.02 seconds, as shown in Table 2. Hanif and Abbas^{9,10} and Abbas et al observed a similar pattern of disintegration time of raft forming tablets containing sodium alginate and pectin. Formulations containing less amount of pectin showed a decrease in disintegration time as compared to the formulations having a high amount of pectin. Higher concentrations of sodium bicarbonate containing formulation rapidly disintegrate as compared to the formulations having a lower amount of sodium bicarbonate.¹¹ The tablet hardness was found between 4.05 ± 0.20 – 4.92 ± 0.12 kg/cm², as shown in Table 2. Akseli et al¹² studied a similar pattern of hardness of tablet formulations using pre- and post-compressional force analysis.

Particle Size Analysis

The particle size of the NCP ranged from 50–500 nm, as mentioned in Figure 1A; 50% of the particles are sized 60–90 nm. The lower percentage of the particles fell in the range of more than 100 nm.¹³ The SEM images of NCP showed particle size in the nano range.¹⁴ The image of SEM showed that the size of the particle was 159 nm, as shown in Figure 1B.

Chemical Properties of Raft

The integrity and strength of the raft depends upon the percentage of NCP in the raft. Antacids are also trapped in the polymeric gel to extend the neutralization time.¹⁰ The antacids in the formulation immediately neutralize the acid in the stomach and reduce the burning sensation. A raft was effectively formed on the surface of the SGF when adding tablets of formulations PPR1 to PPR5, but a raft did not develop when adding formulations PPR6 to PPR9.

The PPR1 formulation contains 85% NCP, the PPR2 formulation contains 78% NCP, the PPR3 formulation contains 62% NCP, as shown in Figure 2A, and the PPR4 formulation contained 61% NCP and the PPR5 formulation contained 94% NCP. The higher concentration of NCP was observed in the PPR5 formulation due to a better crosslinking pattern of raft through calcium ions.¹⁵ The higher concentration of NCP in the raft increases the strength and integrity and effectively stops the reflux of gastric acid from the stomach to esophagus. The calcium ions promote the crosslinking pattern of the nano-structured raft. The raft forming formulations have an initial acid ANC based on their contents of antacid, some of it is utilized to neutralize the acid pocket as soon as the formulation is ingested, however there is a significant residual amount that ends up trapped within the nano-structured raft, as was observed during the process of raft formation. The ANC of the raft was measured to recognize the capacity of the raft to preserve the antacid and deliver a pool of antacids. The raft of the PPR5 formulation showed the highest ANC value (7.0), which was significantly (*P*-value is less than 0.05) different from the other formulations, as shown in Figure 2B. The ANC of the PPR1 formulation was 6.8 (*P*-value is less than 0.05), which was less than the PPR5 formulation but greater than that of the PPR2, PPR3, and PPR4 formulations, which were 3.2, 3.0, and 2.7, respectively. The high ANC of the PPR5 formulation was observed due to the higher concentration of antacids as compared to the other formulations.⁵ Previous studies suggested that antacids are trapped inside the raft, but this has not been measured. A method has been developed to measure the ANC of the raft. Antacids inside the raft will cause initial but often transient neutralization of the acid. An effective formulation for raft formation must have a high ANC and a long neutralization duration. The neutralization profile was used to verify the ability of the raft to neutralize the acid that passes through it. The duration of neutralization of each formulation was determined, and the results showed that the PPR5 formulation showed the longest neutralization duration, ie, 45 minutes. The neutralization durations of the PPR2, PPR3, and PPR4 formulations were 35, 25, and 38 minutes, respectively. The PPR1 formulation was unable to neutralize the acid, as shown in Figure 2C. In 2017, an interesting research article published by Dettmar et al⁵ described the effect of raft structure on the neutralization profile of alginate raft-forming formulations. The PPR5 formulation containing 94% NCP showed a greater

Table 2 Results of Disintegration and Hardness of Tablets

| Code | Disintegration Time (Seconds) | Hardness (kg/cm ²) |
|------|-------------------------------|--------------------------------|
| PPR1 | 54±0.06 | 4.92±0.12 |
| PPR2 | 53±0.02 | 4.81±0.19 |
| PPR3 | 50±0.03 | 4.84±0.14 |
| PPR4 | 56±0.02 | 4.89±0.41 |
| PPR5 | 49±0.06 | 4.90±0.12 |
| PPR6 | 26±0.04 | 4.18±0.55 |
| PPR7 | 23±0.02 | 4.17±0.19 |
| PPR8 | 25±0.01 | 4.13±0.11 |
| PPR9 | 24±0.03 | 4.05±0.20 |

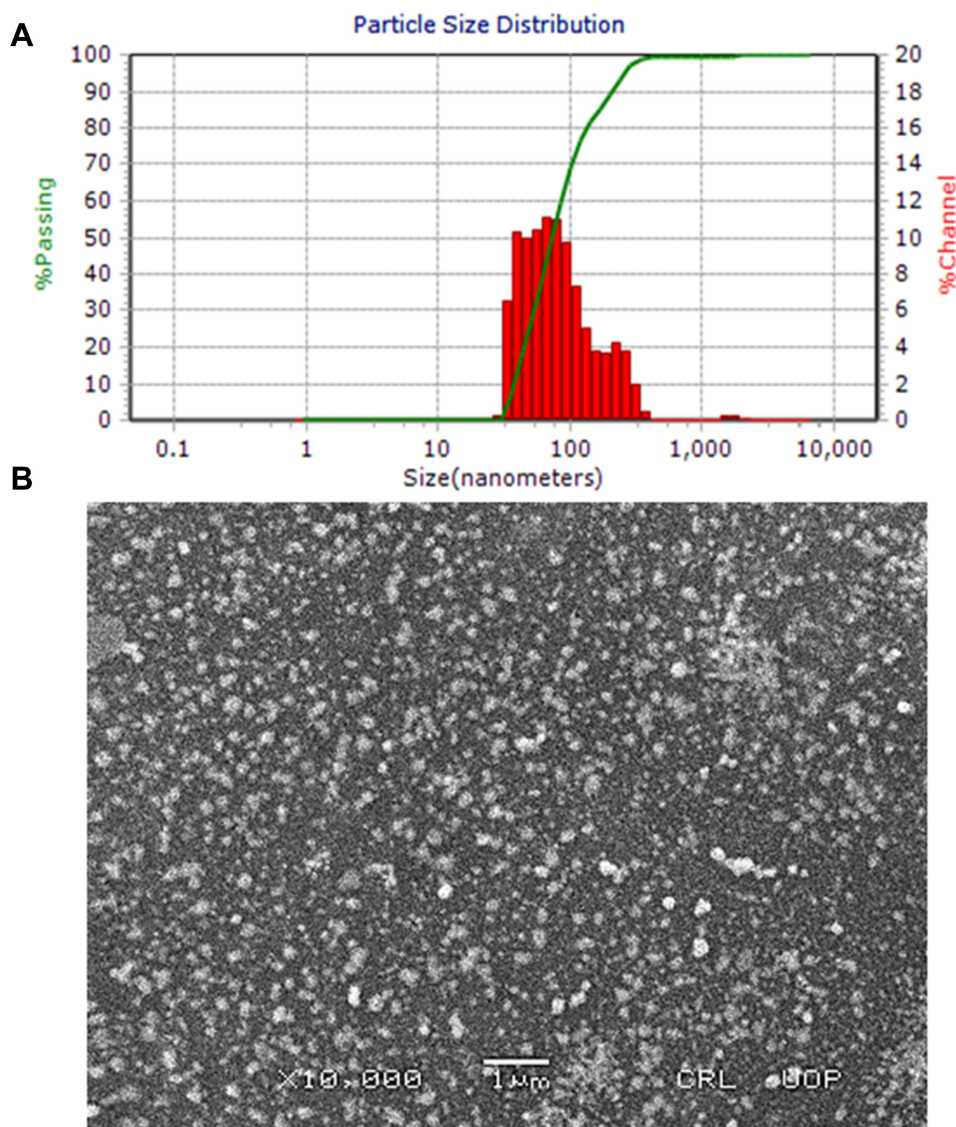


Figure 1 Particle size distribution of NCP prepared for the nanostructured raft (**A**) and SEM images showed the particle size in nano range (**B**).

ANC value and 45 minute duration of neutralization profile as compared to the other formulations. The high ANC and profile of neutralization profile was observed due to the greater concentration of effervescent mixture in the PPR5 formulation. The PPR4 formulation containing a lower percentage of NCP showed a lower value of ANC and moderate profile of neutralization.

Swelling of NCP Raft

The swelling behavior of the NCP raft was evaluated by a gravimetric method in SGF pH 1.2 for 60 minutes, as shown in Figure 2D. The PPR5 formulation raft showed the highest swelling (97% after 60 minutes) in SGF compared to other formulations. This is due to the higher

concentration and gelling behavior of NCP. The -OH of NCP promote wetting and swelling of the raft. The raft of the PPR3 formulation showed the least swelling (60% after 60 minutes) compared to the other formulations. Huanbutta et al⁶ also studied a similar swelling percentage pattern of polymeric dosage forms. The swelling of the raft depends on the concentration of NCP in the prepared formulations. The formulation containing the maximum amount of NCP showed more swelling pattern compared to the formulations containing lower concentration of NCP. The gastric retention of the raft depends upon the swelling of the raft. If the swelling of the raft is more, the retention of the raft in the stomach is more as compared to the raft that has less capacity to swell.

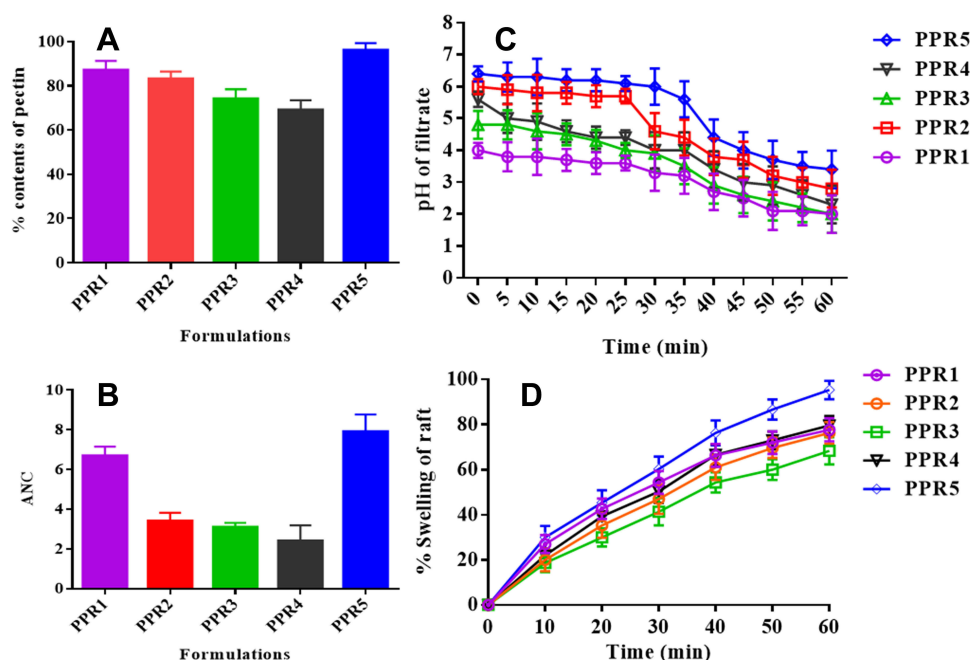


Figure 2 NCP contents (A), ANC (B), neutralization profile (C), and Swelling (D) of nanostructure NCP containing rafts.

Floating Lag Time (FLT) and Total Floating Time (TFT)

The FLT of all formulations ranged from 48–61 seconds, the PPR1 formulation showed the maximum, and PPR4

showed the minimum value of FLT. The TFT of all prepared formulations ranged from 5–6 hours. Abbas et al¹⁰ observed a similar pattern of FLT and TFT of alginate raft forming formulations.

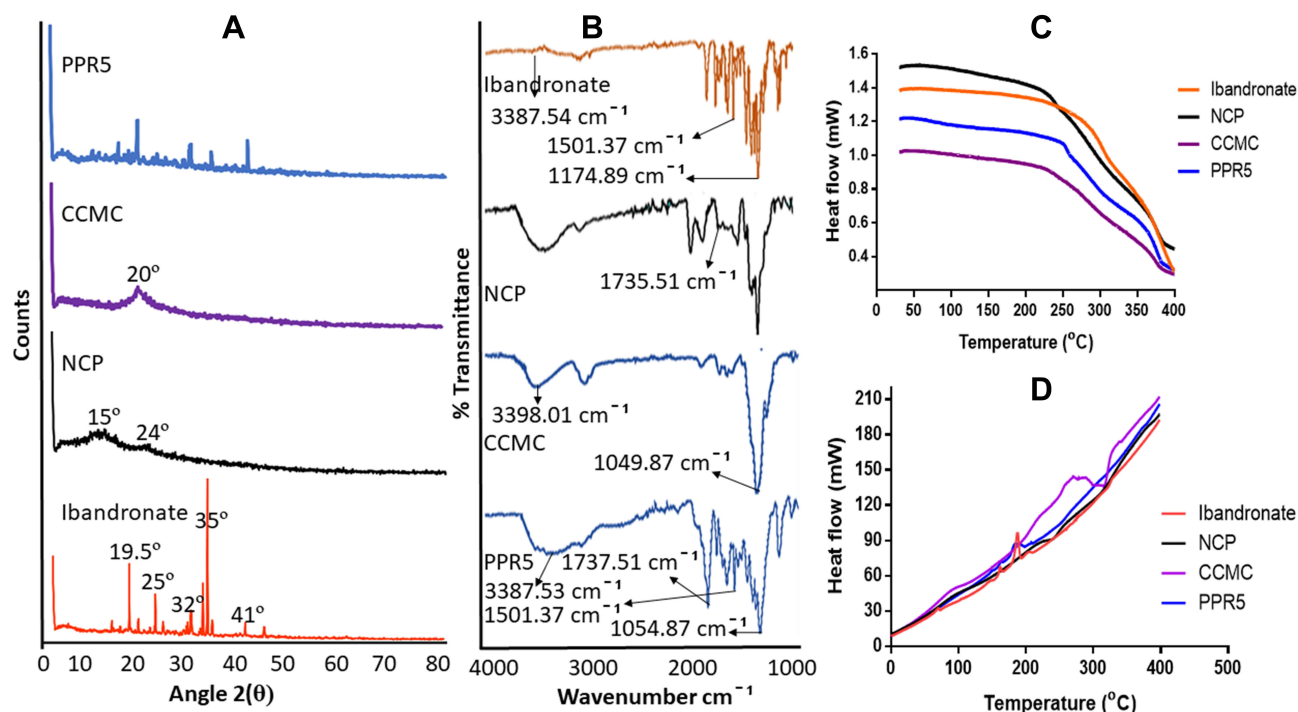


Figure 3 XRD (A), FTIR (B), TGA (C), and DSC (D) of ibandronate, NCP, CCMC, and nanostructured raft of PPR5 formulation.

Characterization

The XRD of ibandronate, NCP, CCMC, and PPR5 formulation are shown in Figure 3A. Ibandronate showed diffraction lines 2(θ) at 17°, 19.5°, 20.5°, 25°, 26°, 32°, 34°, 35°, 41°, and 46° which indicated the crystalline nature of the drug. The diffractograms of NCP showed two diffraction peaks at 15° and 24° 2(θ). The CCMC showed a line of diffraction 2(θ) at 20°. The diffractogram of PPR5 showed many lines of diffraction, but the intensity of peaks was less as compared to the peaks of drug indicated the uniform distribution of drug. The FTIR spectra of the ibandronate, NCP, CCMC, and raft of PPR5 formulation are shown in Figure 3B. The ibandronate showed peaks at 3387.53 cm^{-1} , 1501.37 cm^{-1} , and 1174.89 cm^{-1} due to N-H group stretching, -CH, and C-C group stretching, respectively.¹⁶ The NCP showed a well-defined peak at 1735.51 cm^{-1} due to the stretching of the carbonyl group (C=O) of ester.¹⁷ The CCMC showed two featured peaks at 1049.87 cm^{-1} and 3398.01 cm^{-1} due to vibrations of C-O and O-H groups, respectively.¹⁸ The raft of the PPR5 formulation showed peaks at 3387.53 cm^{-1} , 1737.51 cm^{-1} , 1501.37 cm^{-1} , 1174.89 cm^{-1} , and 1054.87 cm^{-1} , which correspond to the peaks of ibandronate, NCP, and CCMC, respectively, indicating there was no chemical interaction between the drug and polymers. The TGA curves of ibandronate, NCP, CCMC, and PPR5 formulation are shown in Figure 3C. The TGA curves showed only a 5% loss of weight of PPR5 was observed from 40 to 250°C.¹⁹ The DSC thermograms of ibandronate, NCP, CCMC, and the raft

of the PPR5 formulation are shown in Figure 3D. A thermogram of ibandronate showed exothermic peaks at 176°C and 191°C due to loss of crystalline water and 276°C, which showed the melting point of ibandronate.^{20,21} The NCP and CCMC showed endothermic peaks at 250°C and 338°C, respectively. The PPR5 formulation only one exothermic peaks at 191°C indicated the loss of crystalline water and no peak was observed at 276°C, showing the uniform distribution and stability of the ibandronate in the formulation.

Ibandronate Release from Raft

The four different acidic media were used to check the effect of the acid strength on the raft formation and subsequently the release of ibandronate from the optimized raft forming PPR5 formulation. The PPR5 formulation subjected to dissolution for 60 minutes in 0.1 N HCl, 0.5 N HCl, 1 N HCl, and SGF. The PPR5 showed 99.89%, 97.81%, 96.34%, and 96.16% release of ibandronate in SGF, 1 N HCl, 0.5 N HCl, and 0.1 N HCl, respectively, as shown in Figure 4A. Nevertheless, the release rate of ibandronate in various mediums was equivalent, providing over 95% of ibandronate release within 20 minutes. The result indicates that ibandronate mainly existed in the aqueous solution, when the NCP raft was formed on the top layer if in contact with 0.5 N HCl and 1.0 N HCl solution.⁷ The drug was entrapped in the NCP raft when in contact with 0.5 N HCl and 1.0 N HCl solution, the ibandronate seemed to be rapidly diffused out through the foam structures of the NCP raft.

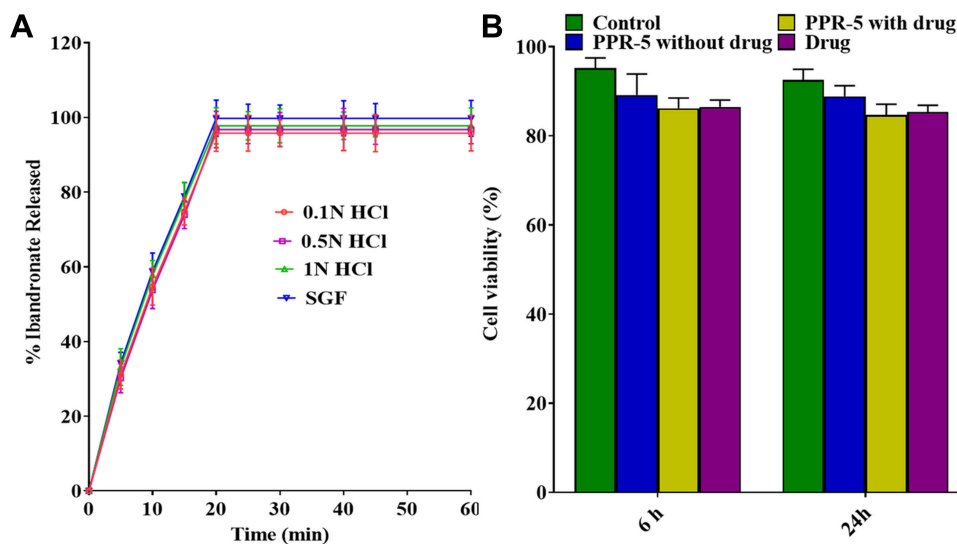


Figure 4 Release profile of ibandronate from nanostructured rafts of PPR5 in different medium (n=6) (A) and cytotoxic study of PPR5 without and with Ibandronate (B).

Formulation (PPR5) Showed Non-Cytotoxic Behavior with Caco-2 Cells

The PPR5 showed up to 90% cell viabilities when Caco-2 cells were treated with or without drug (ibandronate) as well as with drug dispersion compared to control (medium only).²² The cells treated with PPR5-without dispersion showed a viability of 92% and 89% after 6 hours and 24 hours, respectively ($P=0.123$), as shown in Figure 4B. The toxicity profile of PPR5 formulation based on NCP polymer for raft seems to be very close to control, suggesting a safe formulation. Our study is compatible to that of Severino et al,²³ who observed the non-cytotoxic behavior of nanoparticle using Caco-2 and HEPG-2 cell lines.

Stability Studies

At 0-month period the hardness and disintegration time of PPR5 formulation were 4.9 ± 1.09 kg/cm² and 49 ± 2.45 seconds, respectively. The FLT and TFT were 51 ± 2.34 seconds and 5.2 ± 3.12 hours, respectively, as shown in Table 3. The percentage release of drug and assay showed $99.87 \pm 3.09\%$ and $98.54 \pm 1.89\%$ ibandronate, respectively. Goh et al²⁴ observed the similar findings of stability studies of pharmaceuticals containing raft forming polymer. At 1 month, the hardness and disintegration time of tablets were 4.8 ± 1.34 kg/cm² and 47 ± 3.09 seconds, respectively. The FLT and TFT of the raft were 48 ± 3.56 seconds and 5.0 ± 2.67 hours, respectively. The percentage release of drug and assay showed $96.38 \pm 3.01\%$ and $94.38 \pm 2.78\%$ ibandronate, respectively. At the 3rd month, the hardness and disintegration time of tablets were 4.5 ± 1.30 kg/cm² and 43 ± 1.45 , respectively. The FLT and TFT of the raft were 45 ± 1.90 seconds and 4.6 ± 2.87 hours, respectively, as shown in Table 3. The percentage release of drug and assay showed $92.56 \pm 2.98\%$ and $91.29 \pm 3.08\%$ ibandronate, respectively. At the 6th month, the hardness and disintegration time tablets were 4.3 ± 0.23 kg/cm² and 41 ± 2.56 seconds, respectively. The FLT and TFT of the raft were 44 ± 1.51 seconds and 4.4 ± 2.90 hours,

Table 3 Results of Stability Studies

| Duration in Months | Hardness (kg/cm ²) | Disintegration Time (seconds) | FLT (seconds) | TFT (hours) | % Release of Drug | Assay (%) |
|--------------------|--------------------------------|-------------------------------|---------------|----------------|-------------------|------------------|
| 0 | 4.9 ± 1.09 | 49 ± 2.45 | 51 ± 2.34 | 5.2 ± 3.12 | 99.87 ± 3.09 | 98.54 ± 1.89 |
| 1 | 4.8 ± 1.34 | 47 ± 3.09 | 48 ± 3.56 | 5.0 ± 2.67 | 96.38 ± 3.01 | 94.38 ± 2.78 |
| 3 | 4.5 ± 1.30 | 43 ± 1.45 | 45 ± 1.90 | 4.6 ± 2.87 | 92.56 ± 2.98 | 91.29 ± 3.08 |
| 6 | 4.3 ± 0.23 | 41 ± 2.56 | 44 ± 1.51 | 4.4 ± 2.90 | 89.67 ± 3.01 | 89.16 ± 2.45 |

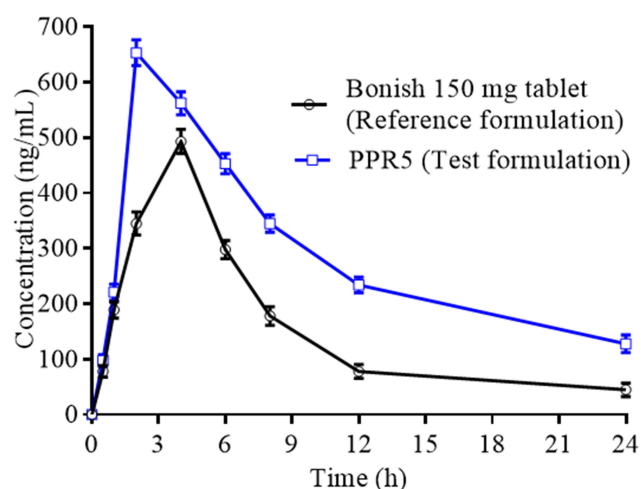


Figure 5 Mean ibandronate profile after oral administration of test and reference formulation (n=6).

respectively. The percentage release of drug and assay showed $89.67 \pm 3.01\%$ and $89.16 \pm 2.45\%$ ibandronate, respectively.

Pharmacokinetic Analysis

To investigate the in vivo behavior of the raft forming ibandronate tablets, the mean plasma concentration-time profiles of ibandronate following oral administration of Bonish tablets (reference formulation) and PPR5 tablet (test formulation) in rats were estimated (Figure 5). Various pharmacokinetic parameters such as C_{max} (ng/mL), t_{max} (h), AUC_{0-t} (ng/mL.h), $AUC_{0-\infty}$ (ng/mL.h), AUMC (ng/mL.h), $t_{1/2}$ (h⁻¹), MRT (h), and kel (h⁻¹) were calculated by non-compartmental approach and presented in Table 4. The difference between the pharmacokinetic parameters was statistically analysed by applying Student's *t*-test (Table 4) at 5% level of significance. From the results, it was observed that the t_{max} for the test formulation was 2 ± 0.025 hours ($P=0.0001$) and that the t_{max} of the reference was 4 ± 1.398 hours ($P=0.0004$). The peak plasma concentration of the PPR5 test and the reference formulation was 653 ± 0.097 ng/mL and 493 ± 0.237 ng/mL, respectively. The $t_{1/2}$ of the

Table 4 Pharmacokinetic Parameters of Ibandronate from PPR5 and Bonish 150 mg Tablets (n=6)

| Parameters | Test Formulation (PPR5) | Reference Formulation (Bonish 150 mg Tablets) | P-value |
|------------------------------|-------------------------|---|---------|
| t_{max} (h) | 2±0.025 | 4±1.398 | 0.0001 |
| C_{max} (ng/mL) | 653±0.097 | 493±0.237 | 0.0002 |
| $t_{1/2}$ (h) | 11.69±2.025 | 6.139±1.142 | 0.0001 |
| $AUC_{(0-t)}$ (ng/mL.h) | 6899.25±3.467 | 3708.25±3.418 | 0.0001 |
| $AUC_{(0-\infty)}$ (ng/mL.h) | 9058.83±5.612 | 4106.81±4.104 | 0.0004 |
| AUMC (ng/mL.h) | 149,523.18±3.676 | 40,051.83±3.412 | 0.0001 |
| MRT (h) | 16.50±1.296 | 9.75±0.379 | 0.0003 |

test and reference formulation was 11.69±2.025 hours and 6.139±1.142 hours, respectively. The observed $AUC_{(0-t)}$ of the PPR5 test formulation was 6899.25±3.467 ng/mL.h, which was higher than the $AUC_{(0-t)}$ of the reference formulation, ie, 3708.25±3.418 ng/mL.h, indicating the bioavailability²⁵ of the PPR5 was higher than that of the reference formulation. The $AUC_{(0-\infty)}$ values of test and reference formulation were 9058.83±5.612 ng/mL.h and 4106.81±4.104 ng/mL.h, respectively. An increase in AUC, an important tool for the estimation of bioavailability, could be correlated to the increase in bioavailability of drug. The use of NCP in the formulation showed the highly porous and absorbent nature of the raft.²⁶ The MRT of the test and reference formulation was 16.50±1.296 hours and 9.75±0.379 hours, respectively, as mentioned in Table 4. The use of NCP enhances the release of ibandronate form PPR5 and PEG 400 improves the penetration of ibandronate through the stomach. The test formulation (PPR5) showed greater bioavailability compared to the reference formulation.

Conclusion

The NCP was successfully prepared from citrus pectin and showed effective and porous raft formation. These newly prepared tablets rapidly dispersed in the SGF, rapidly released the ibandronate, and more than 90% drug was release within 20 minutes. This dosage form effectively neutralizes the acidity of the stomach and maintains the pH of the stomach above 3.5 to prevent the reflux of ibandronate into the esophagus. The bioavailability of newly developed PPR5 was greater than the already marketed formulation. This novel NCP raft forming formulation can be the best alternative for the delivery of ibandronate through the oral route.

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Disclosure

The authors report no conflicts of interest in this work.

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